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*Original*

Efficient management of the water resource in the fresh-cut industry: Current status and perspectives / Manzocco, Lara; Ignat, Alexandra; Anese, Monica; Bot, Francesca; Calligaris, Sonia; Valoppi, Fabio; Nicoli, Maria Cristina. - In: TRENDS IN FOOD SCIENCE & TECHNOLOGY. - ISSN 0924-2244. - 46:2(2015), pp. 286-294. [10.1016/j.tifs.2015.09.003]

*Availability:*

This version is available at: 11381/2919416 since: 2022-03-19T11:12:40Z

*Publisher:*

*Published*

DOI:10.1016/j.tifs.2015.09.003

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# **Efficient management of the water resource in the fresh-cut industry: current status and perspectives**

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## **Abstract**

Among the different food industries, fresh-cut produce manufacturing is one of the major water-intensive, due to the huge consumption of potable water to perform washing operations required to guarantee the safety and quality of the product. Reducing the water footprint of washing is thus a challenge for fresh-cut industries and food researchers. This review paper examines the current status of the water resource management in the fresh-cut industry and critically describes a comprehensive approach to the improvement of the water use efficiency by implementing strategies of water recirculation, reuse and recycling.

**Keywords:** fresh-cut, washing, water saving, decontamination, recycling

## **Introduction**

Water is at the base of humankind's survival and living organisms depend upon it to complete their life cycle and further contribute to natural cycle (Hong-Bo, Li-Ye, Gang, Jin-Heng, & Zhao-Hua, 2007). Issues relevant to population increase, deterioration of surface water quality and climate changes are increasingly requiring to secure water supplies and alleviate environmental loads (EEC, 1991).

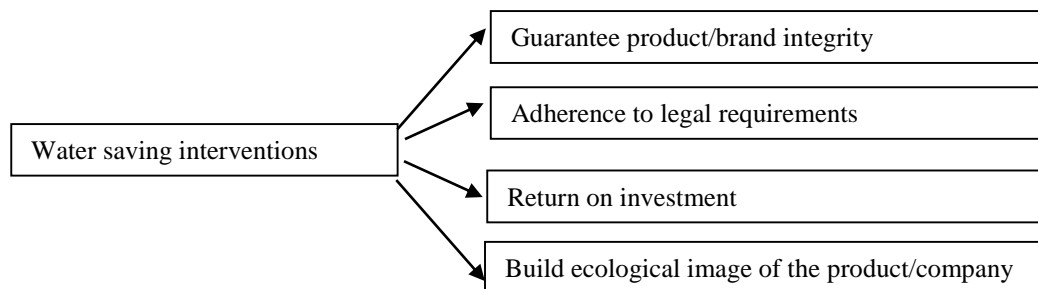
Food production and processing are known to account for the majority of water use globally (Foster et al., 2006). In this sector, the fresh-cut industry is one of the major water-intensive. Water consumption and wastewater volumes are generally in the range of 2-11 m<sup>3</sup>/t and 11-23 m<sup>3</sup>/t of fresh-cut product (FDM-BREF, 2006; Ölmez, 2014). These huge amounts of water are mainly discharged to surface water and make the fresh-cut industry difficult fitting with nowadays global water scarcity. This issue is expected to become particularly critical in the next years, due to the intensification of the demand for fresh-cut produce in developing countries. The minimization of water use and wastewater discharges are thus big challenges for the fresh-cut industry that will be increasingly required to implement sustainable strategies for water saving (Ölmez & Kretzschmar, 2009; Gómez-López, Gobet, Selma, Gil, & Allende, 2013).

By focusing on the eco-efficient management of water, new opportunities and technologies for the environmental performance improvement, that can be also cost-effective, are increasingly under study and possibly applicable for water saving in fresh-cut production. In any case, the actual contribution of these interventions to the sustainable development of fresh-cut vegetable washing strictly depends on the benefits justifying their cost (Fig. 1).

Any innovation allowing washing operations with increased eco-efficiency is required to guarantee, or increase, the safety and quality characteristics of the product in line with industry norms. Yet there must be a return on the investment. Beside tangible profit, non tangible benefits could also come from the opportunity the company may have to build an eco-friendly image. In addition, there is the possibility that some countries will be more specific on the type and amount of certain

chemicals allowed in the waste water discharges and known to be ecologically undesirable. This approach could eventually contribute to justify additional costs involved in water saving interventions.

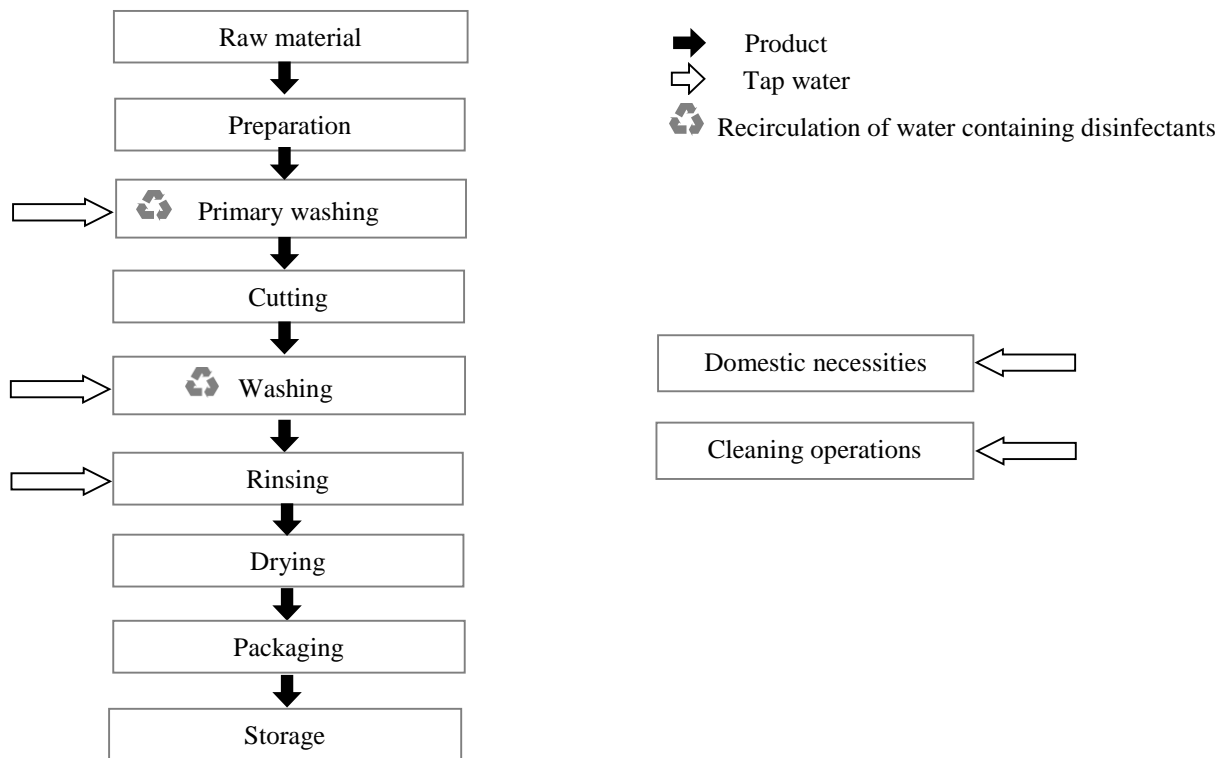
This review paper analyses the current status of the water resource management in the fresh-cut industry, identifying possible strategies for improving water use efficiency and increase the overall sustainability of the production.



**Fig. 1.** Requirements for sustainable water saving intervention.

### **Water management in a typical fresh-cut vegetable production**

Fig. 2 shows the flow of the product in a typical fresh-cut industry. The majority of water is used to perform washing operations, including primary washing to remove gross contamination, a number of consecutive immersions of the product in washing tanks and a final rinse step. Subsidiaries activities requiring water supply are cleaning and sanitizing operations as well as domestic necessities (toilets and staff usage). Current productions generally use water with different properties depending on the nature of the operation to be performed. Water added with chemical disinfectants, such as chlorine and its related compounds, is used to perform the washing steps. The latter include primary washing as well the consecutive passages in washing tanks. The number of passages depends on the organisation of the production flow. By contrast, the final rinsing of vegetables is performed with fresh tap water to remove disinfectant residues. Similarly, tap water is also used for plant cleaning and removal of disinfectants and detergents used to this aim.



**Fig. 2.** Flows of product and water in a typical fresh-cut production process.

### Efficient water management

In the attempt to develop efficient strategies for water saving, the first step is performing a review of the water used within the industry, considering each operation requiring water. This implies a holistic review of what water is actually used for the different applications and the characterisation of effluent water qualities, also in relation to legal requirements. The output is the description of the water flows to/from the production process and represents the water management plan of the industry. By analysing the water management plan, eventual corrective actions for water conservation can be identified and possibly tested. For instance, major savings could be generated by simply controlling leaks or improving on-site cleaning and operating practices. It is evident that the follow-up of any corrective actions should be performed to assess their effectiveness and eventual drawbacks. Following the implementation of the identified water saving interventions, a novel water management plan with improved efficiency is expected to be produced. The main steps for cost-effective use of water resource have been summarised by Williams and Anderson (2006).

It is clear that special attention should be paid when analysing water needs and developing possible water saving interventions potentially applicable to the washing operation. Contrarily to a commonly diffused belief, these huge amounts of water are not required to decrease the vegetable microbial count. The microbial load of vegetables entering the fresh-cut industry may range from 5 to 9 Log units, depending on type of salad, cultivation system, harvesting and handling procedures among other factors (Ölmez & Kretzschmar, 2009; Barth, Hankinson, Zhuang, & Breidt, 2010). Cutting operations, typically performed to produce fresh-cut vegetables, are well known to further increase microbial counts, with effects on both product safety and quality (Ragaert, Devlieghere, & Debevere, 2007). An average reduction of *circa* 1 Log unit in microflora is generally achieved upon washing, due to the sole mechanical removal of microorganisms from the vegetable surface by the water turbulent flow (Allende, Selma, López-Gálvez, Villaescusa, & Gil, 2008). If washing would be performed using tap water only, water would rapidly become highly contaminated, reaching microbial counts in the same order of magnitude of the unwashed salad. Tap water should thus be continuously renewed to avoid microbial proliferation and vegetable cross contamination by spoilage and pathogenic microorganisms. This risk is conventionally controlled by adding water with disinfection chemicals, thus allowing in-tank recirculation of wash water over a longer time. The addition of disinfectants in washing water has thus the sole objective of reducing the overall amounts of potable water required for this operation (Fig. 2).

The overall water print of washing largely depends on the rate of water turnover in the washing tanks. The more effective the disinfectant chemical, the lower the risk of cross-contamination. For this reason, the adoption of an efficient disinfectant could allow the water turnover flow, and thus the overall water consumption, to be significantly reduced. Based on these considerations, there has been a flurry of studies on the efficacy of disinfectant chemicals which are conventionally applied or potentially applicable to control microbial contamination of wash water during vegetable washing. These chemicals do not only include traditional chlorine but also additional chlorine compounds, which seem to exert specific advantages, as well as chlorine-free molecules. Following,

a schematic description of the main criticisms and potentialities of these molecules is reported. A disinfection treatment is regarded as interesting when allowing at least 5 Log reductions on microbial load of wash water. Despite different chemicals reported in Table 1 are expected to achieve this target, no indications about the effect of the selected disinfectant on the water turnover in the washing tanks is available in the literature.

### *Chlorine disinfectants*

Chlorine is the main disinfectant used worldwide because of its potency, low cost and easy use. As a consequence, chlorine and its related compounds are also the chemical oxidants most widely applied to disinfect washing water in the fresh-cut industry. Levels of 50-200 mg/L of free chlorine allow a 6 Log reduction of microbial load of water to be achieved and decrease the concentration of waterborne pathogens below the regulatory limits (EC, 1998; Codex, 2001; Gil, Selma, López-Gálvez, & Allende, 2009). Due to its oxidative potential, chlorinated water can corrode metal surfaces of processing equipment, reducing their overall life. In addition, as already mentioned, chlorine may react with dissolved organic matter to form carcinogenic and/or mutagenic disinfection by-products (DBP), including trihalomethans and haloacetic acid (Richardson & Ternes, 2005; Krasner et al., 2006). Further criticisms of these disinfectants also include toxicity towards operators as well as the necessity of introducing harmful molecules, such as hypochlorite, in food processing plants. A debate is thus on going on whether to ban chlorine and this decision was eventually taken by different European countries (Germany and Switzerland).

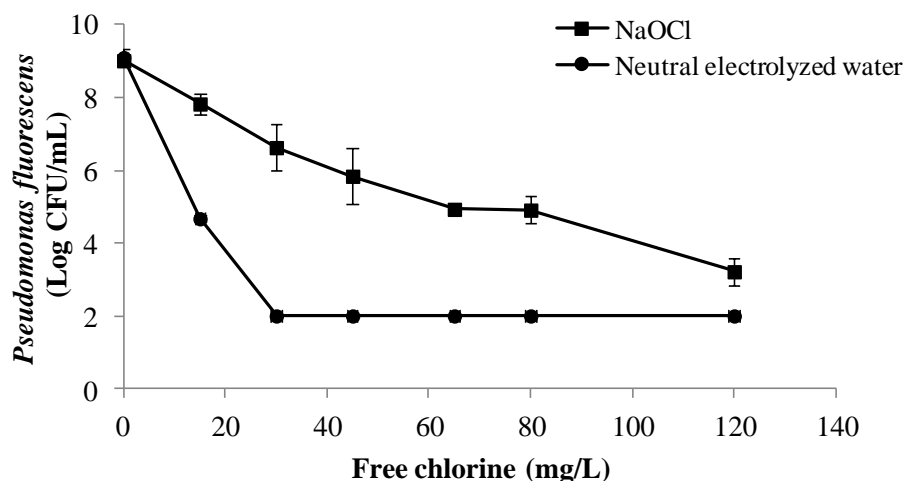
For this reason, any chlorine abuse should be avoided by using the minimum concentration required for the target sanitation level. This implies the knowledge of the chlorine speciation as a function of pH. In water disinfection, gaseous chlorine ( $\text{Cl}_2$ ) and especially sodium hypochlorite ( $\text{NaClO}$ ) are generally used as chlorine sources. Under typical wash water conditions ( $\text{pH} > 4$ ), hypochlorous acid is generated from gaseous chlorine and  $\text{NaClO}$  hydrolysis in water. Hypochlorous acid further dissociates into hypochlorite ( $\text{ClO}^-$ ) and  $\text{H}^+$  (Deborde & von Gunten, 2008). If the water pH is kept

in the range of 6.5-7.5, only hypochlorous and hypochlorite ions are present, while gaseous chlorine concentration is negligible. As hypochlorous acid is the most reactive species against microorganisms, hypochlorite ions would represent a reservoir of HOCl. For this reason, chlorine usage under non-optimal conditions, as occurred in the past, was not only responsible for poor effectiveness in disinfection, but also resulted in an increase of chlorine dose in water up to abuse levels. Nowadays, the decrease in the overall use of chlorine compounds in washing water is simply obtained by their efficient use, following the awareness among operators of the importance of pH measurement and adjustment as well as total, combined and free chlorine concentration determination.

Following the necessity to balance the advantages/disadvantages of chlorination, chlorine based compounds, other than hypochlorite, have also been suggested. In Table 1, the advantages of the main non-hypochlorite alternative compounds (i.e. chloramines, chlorine dioxide and acidified sodium chlorite) as compared to chlorine usage as well as their main drawbacks are shown. More research is certainly required to fully understand their actual efficiency in fresh-cut washing.

More recently, the decrease in chlorine compounds in washing water has also been proposed by exploitation of electrolysed water. It is based on the electrochemical treatments of a diluted (0.5-2 g/L) sodium chloride solution to produce electrolyzed acidic water (pH 2.4-2.8) at the anode and electrolyzed basic solution (pH 11.2-11.6) at the cathode. The neutral electrolysed solution, obtained by mixing these solutions and adjusting pH at *circa* 7.0, is characterised by a redox potential between 600 and 700 mV. Under these conditions, chloride is mainly present as hypochlorous acid, leading to an intense antibacterial effect. Results obtained in the Stayfresh-Ager project has demonstrated that neutral electrolyzed water containing 30 mg/L of free chlorine allows more than 7 Log reductions of *Pseudomonas fluorescens* to be achieved whilst the same result is obtained by conventional chlorination of water with hypochlorite at levels higher than 120 mg/L free chlorine (Fig. 3).





**Fig. 3.** Counts of *Pseudomonas fluorescens* as a function of free chlorine in water solutions added with sodium hypochlorite or produced by electrolysis of a solution containing sodium chloride.

These data suggest that neutral electrolysed solutions could be particularly interesting to decrease the overall presence of chlorine compounds in washing water without reducing the sanitation efficacy (Tomás-Callejas, Martínez-Hernández, Artés, & Artés-Hernández, 2011; López-Gálvez et al., 2012). In other words, the use of neutral electrolysed solutions could guarantee water turnover in washing tanks analogous to those obtained by using hypochlorite, while reducing the overall concentration of chlorine in water. It is noteworthy that solutions are generated on place, making not necessary the introduction of harmful compounds, such as hypochlorite, within the food industry. In addition, these solutions show minimum adverse effects on stainless steel but rapidly loss their antimicrobial activity. To this regard, investigations are need to either increase the solution stability or develop strategies for the renewal of its antimicrobial potential.

Although the trend of maximisation of the sustainable exploitation of the antimicrobial activity of chlorine compounds, their use is still controversial. Chlorine has been shown to fail killing some viruses and microbes, especially in highly contaminated waters. It has also demonstrated to be ineffective in controlling waterborne pathogens such as *Mycobacterium avium* that is ubiquitous in biofilms within water distribution systems (Shannon, Bohn, Elimelech, Georgiadis, Mariñas, &

Mayes, 2008). These aspects, together with the formation of toxic DBP, call for the development of new disinfection strategies based on chlorine combination or substitution with other chemical disinfectants or natural antimicrobials.

### *Chlorine-free disinfectants*

The main chlorine-free disinfection strategies potentially applicable as alternative to the use of chlorine compounds are shown in Table 1. Among these, ozonation is nowadays the most interesting chlorine-free disinfection strategy for washing water. Ozone is a highly unstable molecule which easily forms hydroxyl radicals. This makes it a strong germicidal agent, used for killing pathogenic bacteria since the first industrial ozonation plant for drinking water was built over a century ago. Nowadays it is recognized as GRAS and widely used in Europe and United States. Due to its instability, ozone needs to be generated on-situ by ultraviolet irradiation of an oxygen containing gas or, more diffusely at industrial level, by corona discharge (Kim *et al.*, 2003). Being a gas at 20 °C, it is allowed to solubilise in the water phase by bubble diffusers, injectors or turbine mixers. During water treatment, ozone concentration may range from values lower than 0.1 to 1 mg/L, although higher concentrations can be also obtained under optimal condition. In water, it quickly decompose to form by-products having a very short life (Kim *et al.*, 2003). Being a very strong oxidant, it can show adverse effects on equipment metal surfaces. Several studies have demonstrated the possibility to use ozonised water instead of chlorinated water to perform washing of fresh-cut vegetables (Kim *et al.*, 1999; Beltran, Selma, Marin, & Gil, 2005; Goodburn & Wallace, 2013). A number of different molecules, other than ozone, having oxidative power or characterised by specific antimicrobial activity have been proposed or are currently under study. A detailed description of their potentialities and criticisms is reported in the literature cited in Table 1.

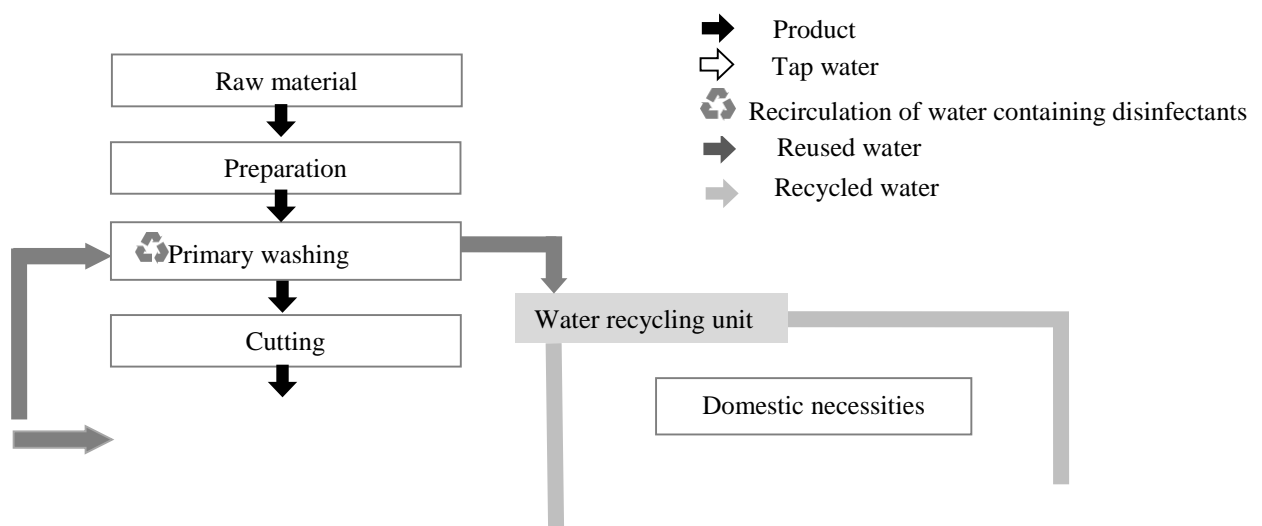
Once the water management plan has been improved by optimising the rate of water turnover in the washing tanks, further water savings can be achieved by implementing strategies of water reuse and

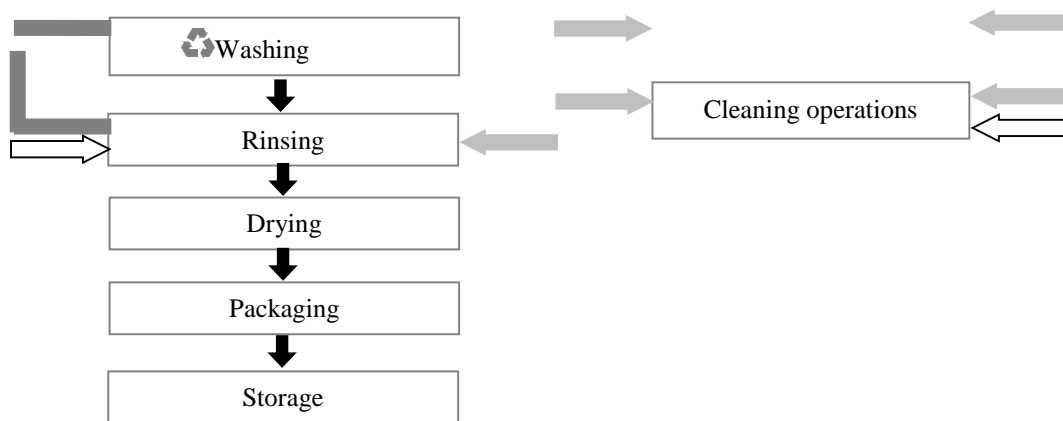
water recycling. Fig. 4 shows, beside the flow of the product, that of water according to these different saving strategies.

### Water reuse

Water reuse is based on the exploitation of the water outflow from a given unit operation to perform another one. In this case, negligible changes in water characteristics are carried out before reuse. An efficient process based on water reuse through the different washing steps has been proposed (Gil *et al.*, 2009). Water is recommended to flow in the opposite direction to product advancement along the different washing steps (Fig. 4). For instance, water from rinsing could be incorporated in the washing tanks, and the latter reused in the pre-washing step. Adequate on-line monitoring of the water characteristics is required to keep the washing efficacy under control.

In this case, the investment costs to implement water reuse are justified in terms of reduced fresh-water costs. Even if the return on the investment is not particularly interesting, the implementation of water reuse could be essential in countries experiencing systematic water shortage which would limit the production itself.





**Fig. 4.** Flows of product and water in a fresh-cut production process with efficient water management.

### Water recycling

Even if water reuse among the washing tanks is performed, a huge amount of wastewater is however produced. It is noteworthy that the largest part of the cost of global wastewater management is pumping and transport from the usage site toward the water depuration treatment site and backwards. Thus, the most efficient solutions of wastewater recovery in any fresh-cut industry should be directly implemented at the production facilities. This concept is at the basis of the development of in-site water recycling strategies. The latter imply specific chemical or physical interventions aiming to modify water properties before re-entering in the production cycle (Fig. 4). Water recycling is more cost-effective than in-tank recirculation and water reuse, since it requires building and setting up of one or more water recycling units. This, in turn, generated the possibility to locally clean wash water and diverge it from municipal wastewater, decreasing the social costs for water treatment and the overall water footprint of the produce. Cleaned water can then be redirected to other internal uses, including not only washing but also plant cleaning and other domestic operations (Fig. 4). When waste water is recycled and intended to re-enter the washing process, microbial disinfection is definitely the main goal. To this regard, it is noteworthy that 5 Log reductions in pathogenic bacteria are generally considered to fulfil the requirement for safe water disinfection and its possible recycling as washing water.

Disinfection of wastewater may be accomplished by different strategies based on the exploitation of physical or chemical stresses, applied alone or in combination, or the physical removal of the contaminating microflora (Table 2).

Although many approaches are virtually exploitable, only a limited number of them actually shows a real applicability to develop water re-circulating units to be implemented in a fresh-cut production line. Among these, the exploitation of light radiation seems very promising. Its antimicrobial effect is due to the ability of ultraviolet light (UV) to damage microbial DNA, blocking DNA transcription and replication thus impairing the cellular functions, eventually leading to cell death (Rame, Chaloupeky, Soikova, & Bencko, 1997). UV light processing is confirmed to be easy to use and characterized by favourable costs of equipment, energy and maintenance (Bintsis, Litoupoulou-Tzanetaki, Robinson, 2000; Guerrero-Beltrán & Barbosa-Cánovas, 2004). It does not leave residues nor forms toxic products (Silva, Lima Filho, Palha, & Sarmento, 2013). In addition, photoreactor design and lamp technology are continuously improving. These aspects led the US Environmental Protection Agency to recognise UV-disinfection as the best current disinfection technology (Hijnen, Beerendonk, & Medena, 2006).

A huge amount of literature data on the germicidal activity of UV radiation is being accumulated since its first application as disinfecting process for drinking water in 1910 (Henry, Helbronner, & Recklinghausen, 1910). Most data refer to kinetics of disinfection of pathogens inoculated in water following its exposure to different doses of UV light (Sommer, Lhotsky, Haider, & Cabaj, 2000; Bintsis *et al.*, 2000). Depending on microorganisms and photoreactor design, a number of Log reductions from 4 to 7 can be easily obtained. Data about the water disinfection by UV-light are not directly applicable for the design of decontamination units for waste water from fresh-cut washing (McKinney, Williams, Boardman, Eifert, & Sumner, 2009; Mounaouer & Abdennaceur, 2012). To this regard, limited literature is available. A 60 min decontamination process of waste water collected from escarole washing in a closed unit was reported to reduce the microbial flora by 4 Log CFU/mL (Selma *et al.*, 2008). The research activity carried out in the Stayfresh-Ager project

allowed demonstrating that higher decontamination levels in total viable count and *Pseudomonas* spp. can be achieved in less than 1 min by optimising the thickness of the waste water layer and the UV light dose during the treatment (Table 3). In particular, following the exposure of a 0.4 cm thick layer of wash water to 0.6 kJ/m<sup>2</sup> UV light, more than 4 Log reductions in total viable count were obtained. These levels of fluence completely inactivated pathogens (i.e. *Salmonella enterica*, *Escherichia coli* and *Listeria monocytogenes*) potentially contaminating wash water. The target value of 5 Log reduction in pathogenic bacteria was reached for the most photoresistant bacteria (*E. coli*) at a UV light dose corresponding to 0.4 kJ/m<sup>2</sup>.

Although highly efficacious, water decontamination by UV light usually requires treatment times ranging from many seconds to min, depending on the photoreactor design and water thickness. This disadvantage could be overcome by using pulsed light, which can be considered an improved way of delivering ultraviolet radiation (Gómez-López, Devleghere, Bonduelle, & Debevere, 2005). It is based on exposure to xenon lamp flashes, which typically last from  $\mu$ s to ms. Light flashes are characterised by an intense broad spectrum of wavelength which includes not only ultraviolet but also visible and infrared light. The latter are known to strengthen the antimicrobial effects of the ultraviolet radiation by local photothermal effect and photophysical disturbance (Dunn, Ott, & Clark, 1995; Guerrero-Beltrán & Barbosa-Cánovas, 2004; Gómez-López *et al.*, 2005; Krishnamurthy, Tewari, Irudayaraj, & Demirci, 2008). Table 3 shows that, by choosing the proper pulsed light fluence (e.g. 11 kJ/m<sup>2</sup>), a rapid disinfection of wash water can be obtained within few ms of treatment. In addition, it was also demonstrated that water recycling in multiple washing cycles significantly affected the spectral properties of wash water due to leaching of chlorophyllian pigments and organic matter (Ignat *et al.*, 2014). However, a significant amount of UV and pulsed light was able to penetrate wash water, yet allowing its efficacious disinfection. Based on this result, recycling up to 5 times of light disinfected wash water did not impair the overall efficacy of salad washing. Although both UV and pulsed light exert interesting potential for developing wash water recycling units, the use of ultraviolet light is nowadays more sustainable from an economic point of

view. However, it is not excluded that technological advances in pulsed light processing could make it increasingly affordable in the next years.

As shown in Table 2, ozonation can also be used for decontamination of wastewater. Although several studies have been performed on the possibility to use ozonised water instead of chlorinated water to perform washing of fresh-cut vegetables (Table 1), to our knowledge no information is available on its use as disinfecting agent of waste water from fresh-cut vegetables. Zimmermann *et al.* (2011) evaluated the disinfection capacity of ozonation in a full-scale municipal wastewater treatment to be 0.5-2.5 Log reduction for *E.coli*. Indeed, the availability on the market of different devices to generate high ozone concentration at reasonable cost certainly focus the operators attention on this technology which is potentially implementable to recycle waste water in a fresh-cut line (Goncharuk *et al.*, 2008; Bialoszewski, Bocian, Bukovska, Czajkowska, Sokół-Leszczyńska, & Tyski, 2010).

Power ultrasound has been suggested as a possible technology to achieve microbial decontamination of wastewater (Piyasena *et al.*, 2003; Blume & Neis, 2004). It is based on the application of ultrasound frequencies higher than 20 kHz, that can be regarded as safe, non-toxic and environmentally friendly (Kentish & Ashokkumar, 2011). Ultrasounds are known to promote cavitation phenomena into the liquid medium with formation of rapidly alternating compression and decompression zones. The latter lead to the development and collapse of small bubbles, thus generating: (i) shock waves associated to local very high temperatures and pressures; (ii) liquid micro-flow; (iii) formation of free radicals and hydrogen peroxide. The efficacy of water disinfection by ultrasound depends on power input, exposure time and microorganism nature (Hulsmans *et al.*, 2010). Literature data indicates that decontamination of water using ultrasound alone does occur but not very rapidly. Inactivation of pathogenic microorganisms is generally reported to be in the 1-2 Log reduction range, difficulty meeting the requirements for water potability. When ultrasound treatments were applied to inactivate *Escherichia coli* O157:H7 in fresh-cut vegetable wash water, a number of Log reduction approaching 5 was only obtained in

treatments time not compatible with industrial processing (60 min) (Elizaquível *et al.*, 2011). Anese (2015) demonstrated that the decontamination efficacy of pathogenic microorganisms (*L. monocytogenes*, *E. coli*, *S. enterica*) in wash water derived from lettuce washing was the result of the contribution of two different effects: the acoustic stress to the microorganism and the ultrasound-induced thermal effect. The latter would largely prevail in the case of heat sensitive microorganisms, such as *E. coli* and *S. enterica*.

When a single technology fails in adequately disinfect waste water, decontamination could however be achieved by combining different processes, thus begetting hybrid technologies. For instance, ultrasounds could be applied in conjunction with other disinfection methodologies, such as light irradiation or ozonation (Mason *et al.*, 2003; Blume & Neis, 2004; Hulsmans *et al.*, 2010). Selma *et al.* (2008) proposed a hybrid process based on water decontamination in a UV-O<sub>3</sub> reactor, demonstrating a microbial reduction of 6.6 Log units after 60 min of water treatment. The disinfection efficacy was found to be higher than that of O<sub>3</sub> and UV applied separately. Paleologu *et al.* (2007) studied the possibility to apply H<sub>2</sub>O<sub>2</sub>-assisted UV/TiO<sub>2</sub> photo catalysis to completely inactivate *E. coli* in wastewater. Disinfection processes can also be combined with membrane separation techniques. The latter are based on the use of microfiltration or ultrafiltration to clean wastewater and redirect it to industrial use. In particular, reverse osmosis followed by UV disinfection or photo catalysis have been claimed to potentially produce water with potable characteristics (Shannon *et al.*, 2008). The development of a hybrid technology, although apparently more cost-effective as compared to individual techniques, may allow the application of each disinfection stress at a lower intensity level, potentially decreasing the operative cost over the long period. The discussion about the actual sustainability of hybrid technologies is the starting point of future research.

## **Conclusions**



There has been much research into the quality and safety of fresh-cut produce. Production techniques, shelf life extension strategies, packaging technologies have mainly concentrate the efforts of the scientists in the last decades. Due to these advancements, produce with high quality standards is now available. Fresh-cut vegetables consumption seem to be stable in industrialised countries but dramatically increasing in developing ones. Is the water footprint of such a production globally sustainable?

If the water consumption issue in industrial vegetable washing will be not carefully considered, a significant increase in the environmental (increase in waste water containing ecologically undesirable chemicals and characterised by high BOD and COD), social (waste water treatment cost) and health (toxicity of water disinfectants and by products) risks is expected in the future.

Fortunately, a recent flurry of activity in water treatment research is offering opportunities in mitigating the impact of this industrial sector. The expectation is that by focusing on the implementation of systems of efficient management of the water resource, sustainable, affordable, safe and robust methods to decrease water consumption in the fresh-cut industry can be developed and implemented. The time required for this to occur will not only depend on the capability to import knowledge and skills from the water purification sector but also on the availability of an adequate normative framework regulating the discharge of water waste.

### **Acknowledgements**

This research was supported by Fondazioni in Rete per la Ricerca Agroalimentare, “AGER - Agroalimentare e Ricerca”. Project 2010 2370.

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## Figure captions

**Fig. 1.** Requirements for sustainable water saving intervention.

**Fig. 2.** Flows of product and water in a typical fresh-cut production process.

**Fig. 3.** Counts of *Pseudomonas fluorescens* as a function of free chlorine in water solutions added with sodium hypochlorite or produced by electrolysis of a solution containing sodium chloride.

**Fig. 4.** Flows of product and water in a fresh-cut production process with efficient water management.

**Table 1. Advantages and limitations of chlorine and-chlorine-free disinfectants, currently used or potentially applicable, in fresh-cut vegetable washing.**

	Molecule	Advantages	Limitations/disadvantages	References
<b>Chlorine compounds</b>	<b>Hypochlorite and related compounds</b>	Easily measured and controlled	Sensory changes	Sanz, Gimenez, Olarte, Lomas, & Portu (2002)
		Low cost and easily available	Formation of unhealthy DBP (THMs, HAAs)	
			pH dependant activity	Baur, Klaiber, Wei, Hammes, & Carle (2005)
			Sensitive to temperature, light, air	Casteel, Schmidt, & Sobsey (2008)
			Efficacy affected by the presence of organic matter	Goodburn and Wallace (2013)
			Corrosive	
			Liberation of chlorine vapours during production	
			Forbidden in some European countries	
	<b>Chloramines</b>	Few unhealthy DBP	Poor biocidal effect	Casani,Rouhany &
		Active for long time	Long contact time	Knochel (2005)
	<b>Chlorine dioxide</b>	Less reactive to organic compounds	Generation on site required	Gómez-López, Rajkovic, Ragaer, Smigc, & Devlieghere (2009)
		Higher antimicrobial activity at neutral pH	Explosive at concentrations > 10% in air	
		Stable over a wide pH range	Not permitted for fresh-cut produce in US and not regulated in EU	Goodburn and Wallace (2013)
		Minimal contact time	Formation of specific DBP	
		High solubility in cold water	Rinsing necessary after washing	
		More stable		
		Less corrosive		
	<b>Acidified sodium chlorite</b>	Greater efficacy (due to low pH)	Little information on DBP produced	Allende, McEvoy, Tao, & Luo, (2009)

				Ramos, Miller, Brandão, Teixeira, & Silva (2013)	
Chlorine-free compounds	Ozone	GRAS	Dependence on organic matter	Kim, Jousef, & Chism (1999)	
		Lower running cost	No residual disinfection	Kim, Jousef, & Khadre (2003)	
		Active at low concentration	activity		
		No pH dependence	High initial investment cost		
		Short contact time	Toxic when inhaled		
		No residues or DBP	Damages to product surface		
			Corrosive		
	Peracetic or Peroxyacetic acid		On-site generation required		
		Efficacy independent on water organic load, temperature and pH	Low antimicrobial efficacy at permitted levels	Artés, Gomez, Aguayo, Escalona, & Artés-Hernandez (2009)	
		Active at low concentrations		Vandekinderen, Devlieghere, De Muelenaer, Ragaert, & Van Camp (2009)	
		Not corrosive (< 80 mg/L)			
		No DBP harmful to human and ecosystem			
		Bromine	Possible synergism with chlorine	Little information on brominated DBP produced	Shere, Kelley, & Richardson (1962)
	Trisodium phosphate	Low corrosivity	High pH (11-12)	Beuchat (1998)	
			Limited antimicrobial efficacy towards <i>Listeria monocytogenes</i>		
	Quaternary ammonium compounds	Colourless, odourless	Cost effective	Aase, Sundheim, Langsrud, & Rørvik (2000)	
		Not corrosive		Parish <i>et al.</i> (2003)	
		Good penetrating ability			
Limited reactivity with organic compounds					

<b>Calcium based compounds</b>	Increase calcium content of the final product	Bitterness and off-flavours if used in association with chlorine	Martín-Diana, Rico, Barry-Ryan, Frías, Mulcahy, & Henehan (2005)
		Limited antimicrobial efficacy	
<b>Hydrogen peroxide</b>	No harmful DBP	Phytotoxic	Akbas and Ölmez (2007)
	Low cost	Impairs product quality: browning/bleaching effect	Alexandre, Brandão, & Silva, (2012)
	Easy to use	Low antimicrobial efficacy at permitted levels	
		Long contact time	
		Requires the removal of H <sub>2</sub> O <sub>2</sub> after processing	
		Potential product toxicity if used in association with silver	
<b>Organic acids (citric lactic, acetic)</b>	Easy to use	Very long exposure time	Akbas and Ölmez (2007)
	No toxicity	Affect product taste and flavour	Ölmez and Kretzschmar (2009)
	GRAS	Relatively low antimicrobial efficacy	
		High COD and BOD values of wastewater	
		Antimicrobial effect dependent on nature of acid and microbial strain	

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DBP: Disinfection by-products.

**Table 2. Main strategies for water disinfection, potentially applicable to wastewater deriving from fresh-cut washing.**

Technology	Principle	Process	Disinfection application	References
Single	Physical stress	UV	Salad wash water	Selma, Allende, López-Gálvez, Conesa, & Gil (2008)
		PL		Ignat, Manzocco, Bartolomeoli, Maifreni, & Nicoli (2014)
				Manzocco, Ignat, Bartolomeoli, Maifreni, & Nicoli (2015)
	Chemical stress	O <sub>3</sub>	Waste water from wine distillery, dairy industries	Gogate and Pandit (2004a)
				Segat <i>et al.</i> (2014)
	Physical and chemical stress	US	Waste water from salad and degradation of pollutants (e.g. chlorinated compounds)	Piyasena, Mohareb, & McKellar (2003)
Hybrid	Physical/chemical stress/separation	UV + O <sub>3</sub>	Wastewater from distillery and tomato industry	Gogate and Pandit (2004b)
		UV + H <sub>2</sub> O <sub>2</sub>		Elizaquivel <i>et al.</i> (2012)
		UV + US		Mason, Joyce, Phull, & Lorimer (2003)
		O <sub>3</sub> + US		Blume and Neis (2004)
		H <sub>2</sub> O <sub>2</sub> + UV/TiO <sub>2</sub>		Hulsmans <i>et al.</i> (2010)
				Gogate and Pandit (2004a)
				Goncharuk, Vakulenko, Shvadchina, Sova, Sitnichenko, & Kalinichenko (2008)
				García-Fernández, Fernández-Calderero,

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		Polo-López, & Fernández-Ibáñez (2015)
RO+ UV	Textile wastewater	Paleologu et al. (2007)
RO + UV/TiO <sub>2</sub>	Water produced in petroleum industry	Shannon <i>et al.</i> (2008)
	Remove toxic compounds from industrial discharge waters	Efligenir, Déon, Fievet, Druart, Morin-Crini, & Crini (2014)
		Dasgupta, Sikder, Chakraborty, Curcio, & Drioli (2015)

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UV: Ultraviolet light

PL: Pulsed light

US: Ultrasonication

O<sub>3</sub>: Ozonation

H<sub>2</sub>O<sub>2</sub>: Addition of hydrogen peroxide

UV-A/TiO<sub>2</sub>: Photocatalysis using titanium dioxide

RO: Reverse osmosis

**Table 3. Log reductions of native and inoculated microorganisms in lamb's lettuce wash water exposed to increasing fluence of UV and pulsed light (modified from Ignat *et al.*, 2014 and Manzocco *et al.*, 2015).**

Light	Fluence (kJ/m <sup>2</sup> )	Microorganisms				
		Native		Inoculated		
		Total viable count	<i>Pseudomonas</i> spp.	<i>S. enterica</i>	<i>L. monocytogenes</i>	<i>E.coli</i>
UV	0.1	3.2	>4.0	>6.0	>7.0	3.7
	0.2	3.6	>4.0	>6.0	>7.0	4.8
	0.4	3.8	>4.0	>6.0	>7.0	5.2
	0.6	>4.0	>4.0	>6.0	>7.0	5.7
	1.2	>4.0	>4.0	>6.0	>7.0	>7.0
Pulsed	2.6	0.5	0.6	0.9	0.6	0.5
	4.4	1.3	1.9	2.9	2.0	3.0
	7.0	2.5	3.6	4.3	4.8	5.6
	11.0	4.9	4.8	>5.0	>6.0	5.4
	17.5	4.9	4.8	>5.0	>6.0	6.3