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Phytochemical characterization of different prickly pear (Opuntia ficus-indica (L.) Mill.) cultivars and botanical parts: UHPLC-ESI-MSnmetabolomics profiles and their chemometric analysis

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1	Phytochemical characterization of different prickly pear (Opuntia ficus-indica (L.)
2	Mill.) cultivars and botanical parts: UHPLC-ESI-MS ⁿ metabolomics profiles and <u>their</u>
3	chemometric analysis
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21 ABSTRACT

Prickly pear is an interesting important source of bioactive compounds. However, a 22 comprehensive characterization of the phytochemical profile of its aerial botanical parts, 23 considering genotypic differences, has not been conducted. This study evaluated the 24 phytochemical composition of four botanical parts (fruit pulp and skin, and young and adult 25 26 cladodes) of six cultivars. Analysis was carried out by using two non-targeted UHPLC-ESI-MSⁿ experimental conditions and assisted with multivariate analysis to facilitate data 27 interpretation. Up to 41 compounds, mainly (poly)phenolic molecules, were identified and 28 quantified, 23 compounds being reported for the first time in Opuntia ficus-indica. Phenolic 29 composition varied significantly depending on the part of the plant. Betalains were detected 30 31 only in the fruit of a red cultivar. This study provided novel insights in terms of identification of bioactives and thorough characterization of botanical parts of prickly pears. This 32 33 information may be used for the development of prickly pear-derived products with high levels of bioactive compounds. 34

35

36 KEYWORDS

37 Cactus; phenolic compounds; metabolomics; foodomics; mass spectrometry; multivariate

38 analysis.

39 1. Introduction

Cactus prickly pear (Opuntia ficus-indica (L.) Mill.) is a plant that could be easily 40 41 cultivated in arid and semiarid climates (Russell & Felker, 1987). It produces edible fruits (called "tuna") and cladodes (fleshy flattened stems, commonly called "nopal"), both used as 42 food and as feed. Prickly pear is employed for nutrition, cosmetic, and ethnopharmacological 43 purposes in the forms of tea, jam, juice, and oil -extracted from the seeds- (Stintzing et al., 44 45 2005). Recently, some authors have highlighted the prospects of different prickly pear aerial parts as good sources of phytochemicals with proven biological activities and high-added 46 value for the food/nutraceutical industry (Barba et al., 2017; Msaddak et al., 2017; Sánchez-47 48 Tapia et al., 2017). This interest in Opuntia bioactives becomes even more relevant when 49 facing upconsidering the need to cope with climate change challenges. Taking into account the tolerance of cactus species to extreme climatic/soil conditions, (Russell & Felker, 1987). 50 the exploitation of its phytochemical content may contribute to its represent a sustainable 51 52 production-activity.

The main phytochemical compounds in prickly pear fruits and cladodes are vitamins, 53 carotenoids, betalains, and (poly)phenolic compounds (Barba et al., 2017; Fernández-López, 54 55 Almela, Obón, & Castellar, 2010; Stintzing et al., 2005). Fruits are good sources of betalains, but the real physiological relevance of these compounds has not been fully unraveled 56 (Moreno, García-Viguera, Gil, & Gil-Izquierdo, 2008). Among the different prickly pear 57 58 phytochemicals, (poly)phenolic compounds are likely those attracting more attention due to their health-related effects (Del Rio et al., 2013; Rodriguez-Mateos et al., 2014; Zanotti et al., 59 2015). The (poly)phenolic fingerprint of prickly pear products is characterized mainly by 60 flavonols and phenolic acids (Fernández-López et al., 2010; Kuti, 2004; Mata et al., 2016; 61 Moussa-Ayoub et al., 2014; Serra, Poejo, Matias, Bronze, & Duarte, 2013; Stintzing et al., 62 63 2005; Yeddes, Cherif, & Trabelsi Ayadi, 2014). However, despite considerable characterizations have been reported (Guevara-Figueroa et al., 2010; Mata et al., 2016;
Moussa-Ayoub et al., 2014; Serra et al., 2013; Yeddes et al., 2014), a detailed profiling of the
bioactive compounds of the aerial parts of prickly pear is lacking.

The accurate characterization of the phytochemical fingerprinting of any vegetal 67 matrix is key to better understand its biological, technological, and nutritional properties 68 69 (Mena et al., 2012). The use of mass spectrometric (MS) metabolomics techniques, assisted 70 by chemometric analysis, has been identified as a valuable asset to evaluate technique in the evaluation of the phytochemical profile of different plant materials rich in bioactive 71 72 compounds (Calani et al., 2013; Eva Ma Sánchez-Salcedo et al., 2016). Analytical approaches allowing easy sample handling and quick, high-throughput chromatographic screening are 73 74 encouraged to accomplish this task (Filigenzi, Ehrke, Aston, & Poppenga, 2011). Nevertheless, the comprehensive study of bioactive compounds may pose some analytical 75 constraints due to the varying capability of diverse chemical scaffolds to respond to the MS 76 ionization settings. Thus, versatile experimental conditions leading to the identification of 77 78 different phytochemical classes are required (Mena et al., 2016).

79 The present work aimed at investigating the phytochemical composition of four 80 different botanical parts (young and adult cladodes, fruit pulp, and skin) of six prickly pear 81 cultivars grown in Spain, extending a preliminary characterization of this plant material (Andreu, Nuncio-Jáuregui, Carbonell-Barrachina, Legua, & Hernández, 2018). The study 82 83 was performed by using two complementary non-targeted UHPLC-ESI-MSⁿ experimental conditions and paired with multivariate analysis to facilitate a comprehensive screening. The 84 high number of samples and the presence of different matrices and classes of phytochemicals 85 represented a major analytical challenge; however, the insights provided in terms of both 86 identification of bioactive compounds and thorough characterization are of interest. 87

88

89 2. Materials and methods

2.1 Chemicals

90

50	
91	Protocatechuic acid, ferulic acid, quercetin-3-O-rutinoside (rutin), naringenin-7-O-rutinoside
92	(narirutin), secoisolariceresinol, and betanin were purchased from Sigma-Aldrich (Steinheim,
93	Germany). HPLC-grade sSolvents were also purchased from Sigma-Aldrich. Water for
94	HPLC analysis was purchased from VWR Chemicals (Fontenay-sous-bois, France).
95	
96	2.2. Plant material
97	Cladodes and fruits of six different cultivars of <i>Opuntia ficus-indica</i> were used for this study.
98	Four cultivars, named "NA", "NT", "NE", and "NO", were collected at the experimental

99 field station of the Miguel Hernandez University in the province of Alicante, Spain

100 (02°03'50''E, 38°03'50''N, and 25 m above sea level). The other two cultivars were

101 collected from private farms in Murcia ("Fresa" cultivar) and Alicante ("Nalle" cultivar) (SE

102 Spain) (less than 50 km far from the experimental station).

103 Young (less than a year) and old cladodes (2 years old), as well as the fruits, were manually harvested during spring and summer of 2015. Ten young cladodes, 10 adult 104 105 cladodes, and 10 fruits from three Opuntia ficus-indica plants per cultivar were harvested. After picking, the plant material was immediately transported to the lab. The spines from the 106 107 cladodes were removed manually, while the fruits were washed under tap water with a brush 108 for 2 minutes. The peels from the fruits were removed manually. The fresh cladodes 109 (young and old), the pulp plus seeds, and the peel were immediately frozen in liquid nitrogen, 110 to be later freeze-dried in an Alpha 2-4 freeze drier (Christ Alpha 2-4; Braum Biotech, 111 Osterode am Harz, Germany) for 24 hours at a pressure reduction of 0.220 mbar. The temperature in the drying chamber was -25 °C, while the heating plate reached 15 °C. 112

113 <u>ThenThereafter</u>, seeds were removed from the pulp, and all the samples were powdered
114 (particle size < 0.4 mm) and packed under vacuum.

115

116 2.3. Extraction of (poly)phenolic compounds

The (poly)phenolic compounds in prickly pear cladodes (young and old) and fruits (pulp and 117 118 skin) were extracted following a protocol previously reported (Sánchez-Salcedo, Mena, García-Viguera, Martínez, & Hernández, 2015). Briefly, 200 mg of freeze-dried powder were 119 mixed with 1 mL of 80% aqueous methanol acidified with formic acid (1%). This mixture 120 121 was then sonicated for 25 min, centrifuged at 10,480 g for 5 min at room temperature, and the supernatant was collected. Two additional extractions were performed for each sample with 122 123 additional 0.5 mL of the extraction solvent, as described above, after which they were centrifuged. The three supernatants were pooled before UHPLC-ESI-MSⁿ analysis. Each 124 125 sample was extracted in triplicate.

126

127 2.4. Liquid chromatography-mass spectrometry (UHPLC-ESI-MSⁿ) analysis

Methanolic extracts of prickly pear parts were analysed using an Accela UHPLC 1250 128 129 equipped with a linear ion trap-mass spectrometer (MS) (LTQ XL, Thermo Fisher Scientific 130 Inc., San Jose, CA, USA) fitted with a heated-electrospray ionization (ESI) probe (H-ESI-II; Thermo Fisher Scientific Inc., San Jose, CA, USA). Separations were performed using a 131 132 XSelect HSS T3 (50_x_2.1 mm), 2.5 µm particle size (Waters, Ireland). Volume injected was 5 µL and column oven was set to 30°C. Two complementary MS experiments were 133 performed, one in negative mode, for non-coloured phenolics, and one using positive 134 ionization, for betalains, following an analytical approach previously developed for the 135 comprehensive identification of (poly)phenolic compounds (Mena et al., 2012). Each sample 136 137 was analysed in duplicate for each experimental condition.

The experimental condition optimized in negative ionization mode for the analysis of 138 non-coloured phenolics was based on the following conditions. The MS worked with a 139 capillary temperature equal to 275 °C, while the source heater temperature was set to 250 °C. 140 The sheath gas flow was 40 units, while both auxiliary and sweep gas were set to 5 units. The 141 source voltage was 3 kV. The capillary and tube lens voltages were -9 and -53 V, 142 143 respectively. Elution was performed at a flow rate of 0.2 mL/min. The gradient started with 144 90% of 0.1% aqueous formic acid and 10% of acetonitrile 0.1% formic acid, followed by a 13-min linear gradient of 10% to 70% acidified acetonitrile. From 13.5 to 14 min the 145 acidified acetonitrile was increased to 80%, followed to 2.5 min of 80% acetonitrile and then 146 4 min at the start conditions to re-equilibrate the column. Analyses were carried out using full 147 scan mode, data-dependent MS³ scanning from m/z 100 to 2000, with collision induced 148 dissociation (CID) equal to 30 (arbitrary units). Pure helium gas was used for CID. 149

For the analysis of betalains, in positive ionization mode, the MS worked with a capillary temperature equal to 275 °C, while the source heather temperature was set to 200 °C. The sheath gas flow was 40 units, while auxiliary gas was set to 5 units, without sweep gas. The source voltage was 4 kV. The capillary voltage and tube lens were 39 and 110 V, respectively. The chromatographic conditions were identical to those used for the previous experimental condition.

Data processing was performed using Xcalibur software from Thermo Scientific. All compounds were identified by comparing with standards, when available, and mass spectral and chromatographic data reported in literature. For quantification purposes, area calculation was performed in selected ion monitoring mode by selecting the relative base peak at the corresponding mass to charge ratio (m/z). The quantification of (poly)phenolics was carried out by comparison with commercial standards, when available. For those compounds that could not be quantified with their corresponding standards, a reference compound was selected based on structural similarity and considering the functional groups that may affect the ionisation properties (i.e., flavonols were quantified as rutin equivalents, lignans as secosiolariceresinol, etc.). Finally, the molecules responding to the ESI source in a unique way with respect to the reference compound of choice, or not reaching the limit of quantification of the corresponding reference compound, were not quantified. Details on the identification and quantification of the phytochemicals are presented in the <u>Supplemental</u> <u>Supplementary</u> Table <u>S</u>1.

170

171 2.5. Statistical analysis

Statistical analyses were performed using the IBM SPSS Statistics 23 software package 172 173 (SPSS Inc., Chicago, IL, USA) and performed at p<0.05 of significance level. Data are 174 presented as mean \pm standard deviation (SD) since the distribution of these variables was 175 normal. A one-way ANOVA with post hoc Tukey HSD test was employed for mean comparisons among cultivars for each botanical part. The assessment of the main effects 176 177 (botanical part, cultivar, and the interaction of botanical part x cultivar) was also carried out with Bonferroni post-hoc tests for multiple comparisons. Principal component analysis 178 179 (PCA) with varimax was performed to explore the differences in the phytochemical profile of 180 the different cultivars and prickly pear parts.

181

182 **3. Results**

183 *3.1. Identification of phytochemicals in Opuntia ficus-indica cladodes and fruits*

The phytochemical screening of prickly pear cladodes (young and old) and fruits (pulp and skin) belonging to six different cultivars was carried out by using two complementary MS experimental conditions. About 120 mass spectra were evaluated for each botanical part, cultivar, experimental condition, and analytical replicate. This exhaustive analysis of the

Opuntia ficus-indica phytochemical composition allowed the tentative identification of up to 188 41 compounds (Table 1). Taking into account the number of compounds identified in prickly 189 190 pear parts, flavonoids were the most relevant class of phytochemicals (16 flavonols, -191 compounds 6, 13, 15, 16, 18, 20-22, 24, 26-28, 31, 32, 37, and 38,- and 2 flavanones, -30 and 33-). Phenolic acids (6 hydroxycinnamic acids, -4, 7, 9, 12, 14, and 36-, 2 phenylpyruvic 192 193 acids, -8 and 35-, 2 hydroxyphenylpropionic acids, -19 and 23-, and 2 hydroxybenzoic acids, 194 -3 and 11-) and lignans (6 compounds, -5, 10, 17, 25, 29, and 34-) were also present. In addition, some other compounds such as betalains (compounds 39-41) and organic acids 195 196 (compounds 1 and 2) were detected.

Two compounds (24 and 39) were identified by comparison with their respective 197 198 analytical standards. Thirty-nine compounds were identified based on their retention time, fragmentation patterns obtained from mass spectra (MS² and MS³ experiments) (Table 1), 199 200 and by comparing their mass spectral characteristic with the available literature (see 201 Supplementary Material, Table S1). The interpretation of the mass spectra fragmentation 202 patterns reported in the literature was not discussed unless of special interest. In this sense, 203 compounds 19, 22, and 26 were tentatively identified according to their characteristic aglycone fragment ions. Compounds 22 and 26 presented a major MS^2 fragment ion at m/z204 205 315 and showed MS³ fragments matching those of other isorhamnetin derivatives (compounds 20, 31, 32, and 37). Compounds 22 and 26 (m/z 755 and 609) also had losses of 206 207 m/z 440 and 294, respectively, which might correspond to sambubioside-rhamnoside and 208 sambubioside moieties; however, the full structure could not be identified and, hence, they were classified simply as isorhamnetin derivatives. Compound 19 presented the same 209 fragmentation pattern of compound 23 and was identified as an isomer of dihydrosinapic 210 acid-hexoside. 23 compounds (3-6, 10-19, 21, 23, 25, 29, 30, 33, 34, 37 and 38) were 211 212 tentatively identified for the first time, as far as we know, in Opuntia ficus-indica.

213

Most of the compounds were identified in all the botanical parts analysed, while some 214 compounds were detected only in some of them (Supplemental-Supplementary Material, Table S1). In the case of betalains, they were only detected in the pulp and skin of the 215 "Fresa" cultivar, the only one presenting an intense red colour. 216

217

218 3.2. Quantification of major (poly)phenolic compounds in Opuntia ficus-indica.

219 The total amount of (poly)phenolic compounds for each botanical part and cultivar is 220 reported at-in Figure 1. There were significant main effects of botanical part, cultivar, and the 221 interaction of botanical part ** cultivar on the content of (poly)phenolic compounds (p<0.001 for all). Regarding the botanical part, the highest (poly)phenolic content was found 222 223 in young cladodes > old cladodes > skin > pulp (p<0.05). Comparison among cultivars for each botanical part showed statistically significant differences on the content of 224 225 (poly)phenolic compounds (Figure 1). The concentration of these compounds varied between 5.3 ("NE") and 14.3 ("Fresa") mg/g dw for young cladodes and from 4.2 ("NO") to 12.4 226 227 ("NE") mg/g dw for old cladodes. The content of (poly)phenolic compounds in fruit skin ranged from 4.3 to 7.1 mg/g dw for "NA" and "NT", respectively, while it varied from 0.7 to 228 5.1 mg/g dw for "NO" and "Nalle", respectively, in fruit pulp. 229

230 The profile of individual (poly)phenolic compounds for each botanical part was dependent on the cultivar (Tables 2-5, Supplementary Figure S1). Twenty-six phenolic 231 232 compounds were quantified in young cladodes, with flavonoids (in particular, 233 flavonols) being the main (poly)phenolic compounds (Table 2). Individual phenolics in young cladodes varied greatly among prickly pear varieties. Myricetin-hexoside (6) was the 234 predominant compound in most of the tested cultivars, except for "NE", where it was present 235 at a very low amount. Young cladodes were also characterized by the presence of relevant 236 237 amounts of some isorhamnetin derivatives (20, 22, and 31), rutin (24), and ferulic acid-

hexoside (9) (Table 2). In the case of old cladodes, up to 25 compounds were quantified 238 239 (Table 3). Similarly to what was reported for young cladodes, flavonols were the major group 240 of (poly)phenolic compounds, and several isorhamnetin glycosides (20, 22, 26, and 31), together with myricetin-hexoside (6) and ferulic acid-hexoside (9), were the main individual 241 242 phenolics (Table 3). Regarding-With respect to fruit skin and pulp, a higher prevalence of 243 phenolic acids over flavonols was noted (Tables 4 and 5). Twenty-six (poly)phenolic 244 compounds were quantified in prickly pear skin, with ferulic acid-hexoside (9), sinapic acidhexoside (12), dihydrosinapic acid-hexoside (23), and isorhamnetin-rutinoside (31) present in 245 high concentrations for most of the cultivars (Table 4). Prickly pear pulp presented a lower 246 number of quantifiable phenolics (21 compounds), the main amount corresponding to a 247 248 ferulic acid derivative (36) (Table 5).

Betalains were not quantified due to the lack of commercially available, pure reference standards (i.e, the purity of the Sigma-Aldrich's betanin and <u>that of</u> other chemical providers <u>is-was</u> not enough to use them as reliable analytical standards, to our concern).

253 3.3. Chemometric classification

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254 Principal component analysis (PCA) was used to better understand the relationships among 255 different botanical parts and cultivars of the species Opuntia ficus-indica in terms of 256 (poly)phenolic composition. Only quantified phenolic compounds (reported in 257 Supplementary Table S1) were taken into account for the PCA. Betalains and organic acids 258 were excluded from this unsupervised multivariate analysis to avoid confounding factors 259 limiting the description of the differences in the (poly)phenolic profile of the samples (i.e., betalains in the pulp and skin of "Fresa" cultivar conditioned strongly the PCA outcomes 260 261 according to preliminary tests).

Two principal components (PCs) were able to explain 61.3% of the total variability. The first PC (PC1), representing 39.3% of the total variance, was positively linked to isorhamnetin derivatives (20, 22, 26, 31, 32, 37), quercetin derivatives (15, 16, 21, 24, 27), kaempferol derivatives (18, 28), and a ferulic acid derivative (7) (Figure 2A), while negatively associated with compounds 10 and 36. PC2 accounted for 22% of the total variance and it was positively correlated with compounds 9, 12, 23, 25, 29, 30, and 38, while it was inversely correlated to compounds 17 and 34 (Figure 2A).

Sample scores for each PC accounted mostly for the similarities among cultivars and 269 270 the differences among botanical parts (Figure 2B). All cultivars presented a similar negative PC1 value for the pulp (low content in flavonoids, rich in lignans), differing only in their 271 272 scores for PC2: "Fresa", "NT", "NA", "NO", and "NE" cultivars formed a sub-cluster with negative scores for PC2, while "Nalle" had positive PC2 values (higher content in phenolic 273 274 acids). For the skin samples, all cultivars displayed neutral scores for PC1 and positive scores for PC2 (medium content in most of the phenolic compounds). "Nalle" cultivar was the skin 275 sample showing a higher value for PC2, characterised by a high content of sinapic acid-276 hexoside (12), dihydrosinapic acid-hexoside (23) and secoisolariciresinol-hexoside (25). 277 278 Most of the cladodes presented similar values for both PCs, although old cladodes had 279 slightly lower PC1 and PC2 scores than young ones. In this sense, young cladodes exhibited 280 a higher <u>flavonol</u> content in flavonols than old cladodes. Nevertheless, some samples showed 281 very high positive scores for PC1, accounting for high concentrations of quercetin and 282 isorhamnetin derivatives, that which was the case of for the old cladodes of "NE" cultivar 283 and the young cladodes of "Fresa".

284

285 4. Discussion

This work investigated the phytochemical profile of four different botanical parts of 286 287 six prickly pear cultivars by using two complementary MS experimental conditions. To the 288 best of our knowledge, this is the first time that so many classes of phytochemicals (betalains, flavonols, flavanones, phenolic acids, lignans, and organic acids) are described in Opuntia 289 ficus-indica, Althoughdespite some accurate works have been found-reported in the literature 290 291 (Guevara-Figueroa et al., 2010; Mata et al., 2016; Moussa-Ayoub et al., 2014; Serra et al., 292 2013; Yeddes et al., 2014), t-This challenging study provideds an exhaustive characterization 293 of the phytochemical profile (betalains, flavonols, flavonols, phenolic acids, lignans, and 294 organic acids) of the aerial parts of Opuntia ficus-indica. Obviously, the range of molecules 295 present in prickly pear phytochemical pool comprises way more than 41 structures, but these 296 may be considered those contributing to a better extent to the definition of its phytochemical fingerprinting, regardless of genotypic differences. From a methodological point of view, this 297 work also reinforces the need for versatile, high-throughput experimental conditions allowing 298 the identification of several groups of bioactives (Filigenzi et al., 2011; Mena et al., 2012; 299 300 Mena et al., 2016; Rak, Fodor, & Abrankó, 2010).

301 While the role of betalains as some of the most interesting phytochemicals in Opuntia 302 genera has been widely discussed for pigmented cultivars during the latest years (Cejudo-303 Bastante, Chaalal, Louaileche, Parrado, & Heredia, 2014; Mata et al., 2016; Stintzing et al., 304 2005), the (poly)phenolic profile of prickly pear has been scarcely assessed. It is known 305 that the concentration of (poly)phenolic compounds in prickly pear depends on genetic and 306 environmental conditions, as well as the part of the cactus plant taken into consideration 307 (Khatabi, Hanine, Elothmani, & Hasib, 2016; Moussa-Ayoub, et al., 2014; Stintzing, et al., 2005). The study of the (poly)phenolic composition of different parts of Opuntia ficus-indica 308 had been previously addressed (Moussa-Ayoub, et al., 2014; Yeddes, et al., 2014). The effect 309 of genotypic differences on the (poly)phenolic profile of prickly pear fruits had also been 310

investigated (Moussa-Ayoub et al., 2014; Stintzing et al., 2005). However, there is a limited 311 knowledge on the (poly)phenolic composition of both edible and residual parts of Opuntia 312 313 taking into account genotypic characteristics (Moussa-Ayoub et al., 2014). This work 314 provides novel insights on-in this regard, with data for individual phenolics on the basis of different botanical parts and genotypes grown under the same environmental conditions. This 315 316 information may be used as starting point for the development of prickly pear-derived 317 products with high levels of (poly)phenolic compounds, as well as for botanical purposes. In addition, the understanding of the phytochemistry of the aerial parts of prickly pear may 318 319 favour an integrated exploitation of cactus orchards.

320 The importance of assessing the (poly)phenolic content of prickly pear fruit pulp 321 is due to their use as edible plants for humans. Since prickly pear fruits are rich in a series of flavonoids and phenolic acids with proven bioactivities (Del Rio et al., 2013; Rodriguez-322 323 Mateos et al., 2014; Zanotti et al., 2015), data on their actual content are key to further 324 explore the biological prospects of prickly pear fruit consumption on human health. The 325 content in (poly)phenolic compounds of the six cultivars was similar in line with previous 326 reports on Opuntia fruits (Moussa-Ayoub et al., 2014; Yeddes et al., 2014), but slightly lower 327 than those recently reported for this same plant material by using a colorimetric method 328 (Andreu et al., 2018). In terms of individual phenolics, the presence of phenolic acids in juice made from pulp has been confirmed (Mata et al., 2016). Regarding flavonols, while some 329 330 authors have identified a few isorhamnetin derivatives in the pulp of Opuntia ficus-indica 331 fruits (Kuti, 2004; Yeddes et al., 2014), others have reported a lack of flavonols in pulp (Moussa-Ayoub et al., 2014). The present characterization accounted for the presence of up 332 to 9 flavonols, as well as several other phenolic scaffolds, in the pulp of prickly pear fruits, 333 which represent a step forward in the definition of the bioactives contained in the main edible 334 335 part of this plant. Although these inconsistencies in the flavonoid profile of prickly pear pulp might be attributed to geographic and genotypic differences, they could likely be due to thesensitivity and accuracy of the methodological approaches used.

338 A higher amount of (poly)phenolic compounds has been reported for fruit skin than for fruit pulp (Moussa-Ayoub et al., 2014; Yeddes et al., 2014), in agreement with our 339 results. Important quantitative differences among cultivars were not found. This similarity 340 341 among cultivars has also been shown for cultivars grown in different countries in terms of 342 flavonol content (Moussa-Ayoub et al., 2014). So far, the (poly)phenolic profile of fruit skins was restricted mainly to flavonols and some phenolic acids (Fernández-López et al., 2010; 343 Kuti, 2004; Mata et al., 2016; Moussa-Ayoub et al., 2014; Serra et al., 2013; Stintzing et al., 344 345 2005; Yeddes et al., 2014), while the present work extends the number of molecules present 346 in this fruit part. Skins, which are usually a waste product, represent a potential source of bioactive compounds that may increase the amounts of (poly)phenolic compounds if used for 347 juice elaboration together with the pulp (Fernández-López et al., 2010; Serra et al., 2013). 348 Considering its phytochemical content not only in phenolics but also in betalains (Stintzing et 349 350 al., 2005), prickly pear fruit skin may also be industrialized for the development of 351 sustainabley alternatives allowing the exploitation of their bioactives as nutraceuticals 352 (Matias et al., 2014; Serra et al., 2013). This would minimize production by-products and 353 might generate profits from a by-product generally lacking economic value.

Cladodes were rich in (poly)phenolic compounds. The (poly)phenolic profile of cladodes had been previously reported to comprise flavonols and phenolic acids (Guevara-Figueroa et al., 2010; Msaddak et al., 2017). The newly-described presence of flavanones and lignans increases the number of bioactive compounds in cladodes and, thus, its interest for human health. Young cladodes exhibited a higher content in (poly)phenolic compounds when compared to their older counterparts, which may be explained by changes in the physiology of the cladode as a consequence of the age and maturation stage (El-Mostafa et al., 2014; Rodríguez-Garcia et al., 2007). Young cladodes are frequently consumed as a green vegetable in salads, sauces, soups, stews, snacks, beverages and desserts in Mexico and Southern US (Stintzing et al., 2005). Therefore, considering their (poly)phenolic content, they may contribute to the total intake of (poly)phenolic compounds with the diet. With respect to old cladodes, their use as a valuable source of bioactives compounds or to produce functional products rich in bioactives should be further explored (Msaddak et al., 2017).

367 From a botanical/evolutionary point of view, the assessment of the (poly)phenolic profile of all the aerial parts of different cultivars of prickly pear represents an important 368 369 advance in the understanding of Opuntia plant biology and defence. Multivariate analysis on prickly pear (poly)phenolic composition accounted for the similarity among cultivars instead 370 371 of among botanical parts, which may indicate the selective synthesis of phenolic scaffolds in 372 each plant part. Among other ecological roles, this fact could be linked to plant defence plant 373 mechanisms, where (poly)phenolic compounds play a key role as antibacterial agents and 374 reducing the palatability and nutrient digestibility for herbivores (Salminen & Karonen, 375 2011).

376 Despite Even though this work contributes significantly to the identification of 377 bioactive compounds in alternative plant sources, a couple of analytical constraints should be 378 acknowledged. The first one is related to betalains. Although the most representative Opuntia betalains were identified only in the only red coloured cultivar (Cejudo-Bastante et al., 2013), 379 380 they were not quantified because of the low purity of the standard-commercially available 381 standard (circa 40%, as stated by the provider). Secondly, an accurate quantification of all the phenolic compounds was not possible due to the unavailability of all their respective 382 reference standards. This led to the semi-quantification of most of the phenolics, which, 383 however, did not impair the conclusions drawn from this study. 384

385

386 5. Conclusions

In summary, this analytical work allowed the characterization of the phytochemical 387 profilesing of four botanical parts from six different prickly pear cultivars. Up to 41 388 compounds, mainly (poly)phenolics, were identified, with being 23 of them being reported in 389 Opuntia ficus-indica for the first time. Moreover, some insights on plant biology with respect 390 391 to phenolic distribution were provided. This information may also be used as starting point 392 for the development of prickly pear-derived products with high levels of (poly)phenolic compounds. Lastly, this analytical approach could also be used in other plant products, 393 394 supposedly rich in phytochemicals.

395

396 SUPPLEMENTAL SUPPLEMENTARY MATERIAL

397 Supplemental Supplementary Table S1. References used for the identification of the
398 phytochemicals described in different *Opuntia ficus-indica* botanical parts (reported in Table
399 1), the compounds used for their quantification, and the occurrence of each compound by
400 botanical part regardless of the cultivar.
401 Supplementary Figure S1. Representative chromatograms for each botanical part of cultivar

- 402 <u>"NT", extracted as base peak chromatogram.</u>
- 403

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515

516 FIGURE CAPTIONS

- 517 Figure 1. Total (poly)phenolic content of the different aerial parts of prickly pear for
- 518 different cultivars, obtained as the sum of individual phenolics. Letters above bars denote
- 519 significant differences at p < 0.05.
- 520 Figure 2. Principal component analysis of different prickly pear aerial parts for six different
- 521 Spanish cultivars. A) loading plot of PC1 versus PC2; B) score plot and distribution of the
- 522 samples in the consensus space. In the loading plot, C# indicates the compound code, as
- 523 reported in Table 1. <u>Non-quantified compounds (1, 2, 4, 5, 8, 11, 13, 14, 19, 35, and 34-41)</u>
- 524 <u>were excluded from the analysis.</u> In the score plot, dark green circles correspond to old
- 525 cladodes, light green ones to young cladodes, red to fruit skin, and orange to fruit pulp.
- 526 "Fresa" cultivar has been abbreviated as "FR", while "Nalle" as "NL".

Table 1. Retention time (RT) and characteristic MS ions of phytochemical compounds identified in different *Opuntia ficus-indica* cultivars and botanical parts.

ы	Compounds	RT [$MS^2(m/r)h$	$MS^{3}(m/7)^{b}$	
Iu.	Compounds	(min)	(<i>m/z</i>)	MIS (<i>m/2</i>)	NIS (<i>m</i> /2) ²	
1	L-Malic acid	1.32	133 a	115 (100), 87 (10)	71 (100), 115 (20)	
2	Citric acid	1.51	191	111 (100), 173 (40)	111 (100), 67 (25)	
3	Protocatechuic acid-hexosidec	1.92	315	153 (100)	109 (100)	
4	Caffeic acid-hexoside ^c	2.69	341	179 (100), 161 (20), 135 (5)	135 (100)	
5	Guaiacyl(8-O-4)ferulic acide	2.80	389	343 (100)	139 (100), 283 (50), 223 (45)	
6	Myricetin-hexoside ^c	3.97	479	317 (100)	179 (100), 151 (45)	
7	Ferulic acid derivative	4.10	517	193 (100), 337 (60), 175 (50)	149 (100), 134 (55), 178 (40)	
8	Piscidic acid	4.18	255	165 (100), 193 (30), 221 (20)	135 (100), 107 (60), 147 (40)	
9	Ferulic acid-hexoside	4.26	355	193 (100), 217 (30), 175 (20)	134 (100), 149 (90), 178 (40)	
10	Guaiacyl(t8-O-4)guaiacyl-hexosidec	4.38	537	375 (100)	327 (100), 195 (50), 179 (20)	
11	Salicylic acid-hexoside ^c	4.42	299	137 (100)	93 (100), 137 (50)	
12	Sinapic acid-hexoside ^c	4.47	385	223 (100)	179 (100), 153 (75), 205 (70),	
13	Ouercetin-malonyl-hexoside ^c	4 51	549	505 (100) 356 (40) 461 (20)	161 (30) 356 (100) 461 (20)	
15		4.51	34)	265 (100), 235 (90), 295 (70),		
14	Ferulic acid-C-hexoside	4.78	355	193 (50)	193 (100), 149 (10)	
15	Quercetin-rhamnose-hexoside-	4.84	755	300 (100), 591 (60), 489 (40)	271 (100), 255 (40), 179 (20),	
		4.00		300 (100), 591 (80), 609 (50),	271 (100), 255 (60), 179 (25),	
16	Rutin-pentoside ^e	4.90	741	475 (45)	151 (20)	
17	Syrinigyl(t8-O-4)guaiacylc	5.03	613	405 (100), 567 (20)	357 (100), 195 (70), 209 (60)	
18	Kaempferol-di-rhamnose-hexoside ^c	5.18	739	575 (100), 285(60), 393 (20)	339 (100)	
19	Dihydrosinapic acid-hexoside isomer ^c	5.20	387	225 (100)	151 (100)	
20	Isorhamnetin- rhamnose-rutinoside	5.25	769	315 (100), 605 (80)	300 (100)	
21	Quercetin-hexoside-pentoside ^c	5.30	595	300 (100), 445 (20), 475 (15)	271 (100), 255 (70), 179 (30), 151 (20)	
22	Isorhamnetin derivative	5.35	755	315 (100), 605 (90), 300 (35), 623 (25)	300 (100)	
23	Dihydrosinapic acid hexoside ^c	5.68	387	255 (100)		
24	Quercetin-3-O-rutinoside (rutin)	5.70	609	301 (100)	179 (100), 151 (60)	
25	Secoisolariciresinol-hexoside ^c	5.71	523	388 (100), 243 (15)	361 (100)	
26	Isorhamnetin derivative	5.75	609	315 (100), 459 (20), 300 (15)	300 (100)	
27	Quercetin-hexoside	5.80	463	301 (100)	179 (100), 151 (60), 257 (20)	
28	Kaampfarol rutinosida	5.08	503	285 (100)	257 (100), 267 (80), 229 (59),	
20	Kaempieroi-rutinoside	5.98	375	285 (100)	241 (50)	
29	Syringaresinol ^c	6.00	417	181 (100), 402 (40), 166 (35)	166 (100)	
30	Naringenin-hexoside ^c	6.02	433	415 (100)	271 (100)	
31	Isorhamnetin-rutinoside	6.09	623	315 (100), 300 (20)	300 (100)	
32	Isorhamnetin-C-hexoside	6.31	477	314 (100), 315 (70), 357 (20), 449 (10)	300 (100), 285 (80), 271 (50)	
33	Naringin ^c	6.33	579	459 (100), 271 (30)	357 (100), 235 (80), 271 (75), 441 (60)	
31	Guaiacyl(8-O-4)syrinigyl(8-	6 20	715	583 (100)	535 (100) 369 (50) 357 (20)	
54	8)guaiacyl-hexosidec	0.38	743	385 (100)	555 (100), 509 (50), 557 (50)	
35	Eucomic acid	7.09	239	179 (100), 149 (80), 221 (20)	107 (100), 151 (20)	
36	Feruloyl derivative	7.15	562	337 (100), 386 (80)	193 (100), 175 (90)	
37	Isorhamnetin pentoside ^c	7.47	447	315 (100)	161 (100)	
38	Trihydroxy-methoxy-flavonol ^c	8.55	315	300 (100)	271 (100), 255 (50)	
Id.	Compounds	RT (min)	$[M]^+$ (m/7)	$MS^2(m/z)$	$MS^3 (m/z)$	
39	Betanin	8.22	551	389 (100)	345 (100), 150 (50), 194 (40)	
40	Proline-betaxanthin	8.37	309	265 (100), 263 (90)	221 (100), 152 (40)	
41	Isobetanin	8.66	551	389 (100)	345 (100), 150 (50), 194 (40)	

^{*a*} MS ions in bold were those subjected to further MS fragmentation. ^{*b*} Abundance relative of each fragment ions is reported in brackets. Compounds 1-38 were identified in negative ionization mode, while compounds 39-41 were detected in positive mode. RT, retention time. ^{*c*} Compounds (tentatively) identified for the first time in *Opuntia ficus-indica*.

 Table 2. Concentration (mg/g dw) of (poly)phenolic compounds in young cladodes of six

 cultivars of *Opuntia ficus-indica*.

Id.	Compounds	FRESA	NA	NALLE	NE	NO	NT
3	Protocatechuic acid- hexoside	$0.09\pm0.03\ a$	$0.03\pm0.01\ b$	$0.05\pm0.02~ab$	$0.02\pm0.00\;b$	$0.07\pm0.01\ ab$	$0.03\pm0.00\ b$
6	Myricetin-hexoside	$4.27\pm0.43\ a$	$2.66\pm0.33~b$	$4.71\pm0.26\ a$	$0.03\pm0.00\ c$	$3.38\pm0.23\ b$	$3.21\pm0.18\ b$
7	Ferulic acid derivative	$0.36\pm0.04\ ab$	$0.36\pm0.03\ ab$	$0.37\pm0.01\ a$	$0.13\pm0.03\ c$	$0.27\pm0.03~\text{b}$	$0.29\pm0.02\ ab$
9	Ferulic acid-hexoside	$0.86\pm0.10\ ab$	$1.19\pm0.13~a$	$0.65\pm0.06\ bc$	$0.31\pm0.16\ c$	$0.81\pm0.10\ b$	$0.96\pm0.11 \text{ ab}$
12	Sinapic acid-hexoside	$0.17\pm0.01\ b$	$0.06\pm0.02~\text{cd}$	$0.02\pm0.01\ d$	$0.47\pm0.03~a$	$0.02\pm0.02\ d$	$0.10\pm0.02~\text{c}$
15	Quercetin-rhamnose- hexoside-rhamnose	$0.15\pm0.01\ a$	$0.09\pm0.01\ b$	$0.04\pm0.00\ c$	$0.05\pm0.01~\text{c}$	$0.04\pm0.01\ c$	$0.08\pm0.01\ b$
16	Rutin-pentoside	$0.10\pm0.02\ a$	$0.06\pm0.03\ ab$	$0.03\pm0.01\ b$	$0.08\pm0.02\ ab$	$0.04\pm0.01\ b$	$0.09\pm0.00\ ab$
17	Syrinigyl(t8-O- 4)guaiacyl	$0.15\pm0.02\ a$	$0.06\pm0.01~\text{cd}$	$0.10\pm0.03~bc$	$0.03\pm0.00\;d$	$0.12\pm0.02 \text{ ab}$	$0.03\pm0.01\;d$
18	Kaempferol-di- rhamnose-hexoside	$0.47\pm0.13\ ab$	$0.34\pm0.02\ ab$	$0.49\pm0.08\ ab$	$0.08\pm0.02~\text{c}$	$0.53\pm0.07\;a$	$0.31\pm0.05\ b$
20	Isorhamnetin- rhamnose-rutinoside	$0.82\pm0.06\;a$	$0.58\pm0.07\ b$	$0.20\pm0.02\ c$	$0.58\pm0.10\ b$	$0.29\pm0.06\ c$	$1.00\pm0.12\ a$
21	Quercetin-hexoside- pentoside	$0.12\pm0.02\ a$	$0.06\pm0.01\ b$	$0.03\pm0.01\ b$	$0.05\pm0.00\ b$	$0.03\pm0.01\ b$	$0.04\pm0.00\;b$
22	Isorhamnetin derivative	$0.62\pm0.04\ ab$	$0.39\pm0.07\ bc$	$0.20\pm0.02\ c$	$0.75\pm0.19\ a$	$0.29\pm0.06\ c$	$0.84\pm0.08\ a$
23	Dihydrosinapic acid hexoside	$0.11\pm0.01~\text{b}$	$0.06\pm0.01~\text{c}$	$0.04\pm0.00\ cd$	$0.16\pm0.01\ a$	$0.02\pm0.00\ d$	$0.07\pm0.01~\text{c}$
24	Quercetin-3- <i>O</i> - rutinoside (rutin)	$1.80\pm0.29\;a$	$0.61\pm0.23\ b$	$0.41\pm0.09\;b$	$0.21\pm0.04\ b$	$0.46\pm0.06\ b$	$0.40\pm0.03\ b$
25	Secoisolariciresinol- hexoside	-	-	$0.02\pm0.00\ a$	-	-	$0.01\pm0.00\ b$
26	Isorhamnetin derivative	$0.43\pm0.06~b$	$0.31\pm0.04\ bc$	$0.17\pm0.03~\text{c}$	$0.62\pm0.01\ a$	$0.23\pm0.02~\text{c}$	$0.64\pm0.09\ a$
27	Quercetin-hexoside	$1.02\pm0.62\ a$	$0.57\pm0.15\ ab$	$0.28\pm0.06\ ab$	$0.06\pm0.01\ b$	$0.29\pm0.03\ ab$	$0.23\pm0.03\ b$
28	Kaempferol- rutinoside	$0.77\pm0.07\ a$	$0.23\pm0.03~\text{c}$	$0.46\pm0.01\ b$	$0.22\pm0.03~\text{c}$	$0.41\pm0.03\ b$	$0.43\pm0.00\ b$
29	Syringaresinol	$0.17\pm0.02\ a$	$0.03\pm0.01\ b$	$0.05\pm0.01\ b$	$0.03\pm0.01\ b$	$0.04\pm0.00\ b$	$0.04\pm0.01\ b$
30	Naringenin-hexoside	$0.05\pm0.01\ a$	$0.05\pm0.01\ ab$	$0.03\pm0.01\ c$	$0.05\pm0.01\ ab$	$0.03\pm0.01~\text{bc}$	$0.03\pm0.00\ bc$
31	Isorhamnetin- rutinoside	$0.94\pm0.05\ b$	$0.56\pm0.11\ c$	$0.31\pm0.01\ d$	$1.22\pm0.10\ a$	$0.40\pm0.08~\text{cd}$	$0.93\pm0.08\ b$
32	Isorhamnetin-C- hexoside	$0.61\pm0.08\;a$	$0.46\pm0.06\ b$	$0.24\pm0.03~\text{c}$	$0.07\pm0.01\;d$	$0.25\pm0.06\ c$	$0.19\pm0.02\ cd$
33	Naringin	$0.04\pm0.01\ ab$	$0.03\pm0.01\ b$	$0.05\pm0.00\ a$	$0.03\pm0.00\ b$	$0.01\pm0.00\ c$	$0.03\pm0.01\ ab$
34	Guaiacyl(8- <i>O</i> - 4)syrinigyl(8- 8)guaiacyl-hexoside	$0.05\pm0.01\ a$	$0.03\pm0.00\ ab$	$0.03 \pm 0.00 \text{ ab}$	$0.02\pm0.00\ b$	$0.02\pm0.01\ ab$	$0.01\pm0.00\ b$
37	Isorhamnetin pentoside	$0.08\pm0.01\ a$	-	-	-	-	$0.05\pm0.00\ b$
38	Trihydroxy-methoxy- flavonol	$0.05\pm0.01\ a$	0.02 ± 0.00 bcd	$0.02\pm0.01~\text{bc}$	$0.01\pm0.00\ cd$	$0.01\pm0.00\ d$	$0.03\pm0.01\ ab$

 Table 3. Concentration (mg/g dw) of (poly)phenolic compounds in old cladodes of six

 cultivars of Opuntia ficus-indica.

ы	Compounds	EDESA	NA	NALLE	NE	NO	NT
Iu.	Compounds	FRESA	INA	NALLE	NE	NO	N1
3	hexoside	$0.02 \pm 0.01 \text{ b}$	$0.02\pm0.00\ b$	$0.06\pm0.01\ a$	$0.03\pm0.01\ b$	$0.03\pm0.00\ b$	$0.01\pm0.02~b$
6	Myricetin-hexoside	$0.76\pm0.13\ b$	$0.03\pm0.00\;d$	$0.61\pm0.12\ bc$	$2.43\pm0.06\ a$	$0.39\pm0.09\ c$	$0.79\pm0.10\ b$
7	Ferulic acid derivative	$0.33\pm0.01\ a$	$0.20\pm0.09~b$	$0.37\pm0.04\ a$	$0.33\pm0.03\ a$	$0.28\pm0.04\ b$	$0.37\pm0.02\ a$
9	Ferulic acid-hexoside	$1.82\pm0.16\ a$	$1.13\pm0.21~\text{bc}$	$1.27\pm0.16~\text{b}$	$0.81\pm0.12~\text{cd}$	$0.41\pm0.05~\text{e}$	$0.50\pm0.04\ de$
12	Sinapic acid-hexoside	$0.30{\pm}~0.05~ab$	$0.32\pm0.07\ a$	$0.11\pm0.02~\text{cd}$	$0.06\pm0.01\ d$	$0.30\pm0.04\ ab$	$0.19\pm0.02\ bc$
15	Quercetin-rhamnose- hexoside-rhamnose	$0.07\pm0.01\ ab$	$0.04\pm0.01\ bc$	$0.02\pm0.01~\text{c}$	$0.07\pm0.01\ a$	$0.03\pm0.01\ c$	$0.07\pm0.01\ ab$
16	Rutin-pentoside	$0.07\pm0.01\ a$	$0.03\pm0.01\ b$	$0.02\pm0.01\ b$	$0.06\pm0.01\ a$	$0.03\pm0.00\;b$	$0.08\pm0.01\ a$
17	Syrinigyl(t8-O- 4)guaiacyl	$0.03\pm0.00\;b$	$0.02\pm0.01\ b$	$0.02\pm0.00\;b$	$0.21\pm0.06\ a$	$0.02\pm0.00\ b$	$0.02\pm0.00\ b$
18	Kaempferol-di- rhamnose-hexoside	$0.10\pm0.02\ bc$	$0.05\pm0.01\ bc$	$0.04\pm0.02\ c$	$0.38\pm0.08\ a$	$0.15\pm0.02\;b$	$0.10\pm0.01\ bc$
20	Isorhamnetin- rhamnose-rutinoside	$1.08\pm0.18\;a$	$0.54\pm0.10\ b$	$0.38\pm0.07\ b$	$0.35\pm0.04\ b$	$0.48\pm0.06\ b$	$1.16\pm0.10\ a$
21	Quercetin-hexoside- pentoside	$0.04\pm0.01\ bc$	$0.03\pm0.00\ bc$	$0.02\pm0.01\ c$	$0.06\pm0.00\ a$	$0.02\pm0.00\ c$	$0.04\pm0.00\ ab$
22	Isorhamnetin derivative	$0.76\pm0.12\ b$	$0.43\pm0.07\ c$	$0.40\pm0.26\ c$	$0.38\pm0.08\ a$	$0.40\pm0.09\ c$	$0.89\pm0.06\ bc$
23	Dihydrosinapic acid hexoside	$0.16\pm0.03\ bc$	$0.11\pm0.01\ b$	$0.28\pm0.07\ a$	-	$0.21\pm0.04\ ab$	$0.11\pm0.01\ b$
24	Quercetin-3- <i>O</i> - rutinoside (rutin)	$0.34\pm0.05\ b$	$0.09\pm0.02~\text{c}$	$0.05\pm0.05\ \text{c}$	$1.66\pm0.16\ a$	$0.10\pm0.01~\text{c}$	$0.15\pm0.00\ bc$
25	Secoisolariciresinol- hexoside	-	-	$0.01\pm0.00\ a$	-	$0.01\pm0.00\ a$	$0.01\pm0.00\ a$
26	Isorhamnetin derivative	$0.62\pm0.07~b$	$0.42\pm0.02\ b$	$0.30\pm0.04\ b$	$1.88\pm0.45\ a$	$0.29\pm0.04\ b$	$0.74\pm0.09\;b$
27	Quercetin-hexoside	$0.22\pm0.04\ b$	$0.04\pm0.02~b$	$0.01\pm0.00~b$	$1.61\pm0.29\ a$	$0.04\pm0.00\ b$	$0.05\pm0.00\ b$
28	Kaempferol- rutinoside	$0.15\pm0.04\ bc$	$0.05\pm0.01\ c$	$0.07\pm0.01~\text{c}$	$0.75\pm0.08\ a$	$0.20\pm0.02\ b$	$0.23\pm0.01\ b$
29	Syringaresinol	$0.06\pm0.02\ a$	-	$0.04\pm0.01\ ab$	-	$0.03\pm0.01\ b$	-
30	Naringenin-hexoside	$0.06\pm0.02\ a$	$\begin{array}{c} 0.02\pm0.03\\ abc \end{array}$	-	$0.01\pm0.00\ c$	$0.01\pm0.00\ bc$	$0.04\pm0.01\ ab$
31	Isorhamnetin- rutinoside	$1.19\pm0.13\ a$	$0.66\pm0.08\ c$	$0.16\pm0.03\ c$	$0.73\pm0.06\ b$	$0.72\pm0.13\ b$	$1.27\pm0.12~a$
32	Isorhamnetin-C- hexoside	$0.09\pm0.03\ b$	$0.03\pm0.01~\text{bc}$	$0.01 \pm 0.00 \text{ c}$	$0.50\pm0.07\ a$	$0.01\pm0.00\ bc$	$0.08\pm0.02~\text{bc}$
33	Naringin	$0.03\pm0.01\ a$	$0.03\pm0.00\ ab$	$0.02\pm0.01\ ab$	$0.01\pm0.00\ ab$	$0.02\pm0.00\ ab$	$0.03\pm0.00\ a$
34	Guaiacyl(8- <i>O</i> - 4)syrinigyl(8- 8)guaiacyl-hexoside	$0.03\pm0.00\ a$	$0.02\pm0.01~\text{a}$	$0.04\pm0.02~a$	$0.03\pm0.01\ a$	$0.03\pm0.01\ a$	$0.02\pm0.00\ a$
38	Trihydroxy-methoxy- flavonol	$0.02\pm0.00\ a$	$0.01\pm0.00\ a$	$0.02\pm0.02\ a$	$0.03\pm0.01\ a$	$0.02\pm0.01\ a$	$0.03\pm0.01\ a$

Table 4. Concentration (mg/g dw) of (poly)phenolic compounds in fruit skin of six cultivars

of Opuntia ficus-indica.

	a .	EDEG (NO	N/T
ld.	Compounds	FRESA	NA	NALLE	NE	NO	NT
3	Protocatechuic acid- hexoside	$0.01\pm0.00\;b$	$0.03\pm0.01 \ ab$	$0.08\pm0.04\ a$	$0.02\pm0.00 \text{ ab}$	$0.07\pm0.02\ a$	$\begin{array}{c} 0.02 \pm 0.001 \\ ab \end{array}$
6	Myricetin-hexoside	$0.02\pm0.00\ c$	$0.01\pm0.00\ c$	$0.03\pm0.01\ \text{c}$	$0.01\pm0.00\ c$	$0.08\pm0.02\ b$	$0.56\pm0.04\ a$
7	Ferulic acid derivative	$0.23\pm0.06\ b$	$0.15\pm0.03\ b$	$0.37\pm0.07\ a$	$0.23\pm0.02\ b$	$0.23\pm0.03\ b$	$0.39\pm0.02\ a$
9	Ferulic acid-hexoside	$1.55\pm0.22\ ab$	$1.03\pm0.15~\text{bc}$	$1.03\pm0.32~\text{bc}$	$0.82\pm0.20\ c$	$1.16\pm0.15~\text{bc}$	$1.81\pm0.28\;a$
10	Guaiacyl(t8-O- 4)guaiacyl-hexoside	-	-	-	-	-	$0.02\pm0.00\ a$
12	Sinapic acid-hexoside	$0.47\pm0.08\ b$	$0.62\pm0.13\ b$	$1.72\pm0.41\ a$	$0.81\pm0.11\ b$	$0.64\pm0.08\ b$	$0.47\pm0.09\ b$
15	Quercetin-rhamnose- hexoside-rhamnose	$0.03\pm0.01\ ab$	$0.02\pm0.00\ ab$	$0.01\pm0.00\ b$	$0.02\pm0.00\ ab$	$0.01\pm0.00\ b$	$0.03\pm0.01\ a$
16	Rutin-pentoside	$\begin{array}{c} 0.04 \pm 0.02 \\ abc \end{array}$	$0.02\pm0.01\ c$	$0.03\pm0.00\ bc$	$0.06\pm0.01\ a$	$0.02\pm0.00\ c$	$0.05\pm0.01\ ab$
17	Syrinigyl(t8-O- 4)guaiacyl	$0.03\pm0.01\ a$	$0.01\pm0.00\ bc$	$0.03\pm0.00\ ab$	$0.03\pm0.00\ a$	-	$0.03\pm0.00\ a$
18	Kaempferol-di- rhamnose-hexoside	$0.01\pm0.00\ ab$	$0.01\pm0.00\ b$	$0.02\pm0.00\ a$	$0.02\pm0.00\ ab$	$0.03\pm0.00\ a$	$0.02\pm0.00\ ab$
20	Isorhamnetin- rhamnose-rutinoside	$0.45\pm0.08\ ab$	$0.28\pm0.05\ bc$	$0.26\pm0.04\ bc$	$0.34\pm0.03\text{ bc}$	$0.23\pm0.01\ c$	$0.61\pm0.15~a$
21	Quercetin-hexoside- pentoside	$0.02\pm0.01\ b$	$0.02\pm0.00\ ab$	$0.02\pm0.00\ b$	$0.04\pm0.01\ a$	$0.01\pm0.00\ b$	$0.02\pm0.01\ b$
22	Isorhamnetin derivative	$0.42\pm0.07\ bc$	$0.31\pm0.08\ b$	$0.44\pm0.03\ bc$	$0.72\pm0.07\ a$	$0.38\pm0.02\ b$	$0.65\pm0.15\ ab$
23	Dihydrosinapic acid hexoside	$0.35\pm0.08\ c$	$0.55\pm0.09\ cd$	$1.16\pm0.16\ a$	$0.93\pm0.11 \text{ ab}$	$0.66\pm0.08~bc$	$0.54\pm0.13\ cd$
24	Quercetin-3- <i>O</i> - rutinoside (rutin)	$0.10\pm0.01~\text{bc}$	$0.10\pm0.02\;bc$	$0.06\pm0.01\ c$	$0.16\pm0.04\ ab$	$0.08\pm0.01\ c$	$0.18\pm0.03\ a$
25	Secoisolariciresinol- hexoside	-	$0.03\pm0.00\ bc$	$0.13\pm0.04\ a$	$0.02\pm0.00\ c$	$0.08\pm0.01\ b$	-
26	Isorhamnetin derivative	0.30 ± 0.06 abc	$0.27\pm0.05\ bc$	$\begin{array}{c} 0.33 \pm 0.04 \\ abc \end{array}$	$0.49\pm0.09\ a$	$0.21\pm0.03~\text{c}$	$0.44\pm0.11\ ab$
27	Quercetin-hexoside	$0.07\pm0.02\ a$	$0.04\pm0.01\ bc$	$0.02\pm0.00\ c$	$0.06\pm0.00\ ab$	$0.02\pm0.01\ c$	$0.08\pm0.01\ a$
28	Kaempferol- rutinoside	$0.04\pm0.01\ bc$	$0.02\pm0.00\ c$	$0.06\pm0.01\ ab$	$0.05\pm0.00\;b$	$0.07\pm0.01\ a$	$0.06\pm0.01\ ab$
29	Syringaresinol	$0.20\pm0.03\ a$	$0.11\pm0.02\ b$	$0.13\pm0.04\ b$	$0.13\pm0.00\;b$	$0.12\pm0.01\ b$	$0.24\pm0.02\ a$
30	Naringenin-hexoside	$0.06\pm0.02~\text{ab}$	$0.02\pm0.01\ b$	$0.18\pm0.05\ a$	$0.12\pm0.01\ ab$	$0.07\pm0.01\ ab$	$0.07\pm0.01\ ab$
31	Isorhamnetin- rutinoside	$0.53\pm0.12\ b$	$0.53\pm0.10\ b$	$0.61\pm0.04\ ab$	$0.85\pm0.19\ a$	$0.58\pm0.03\ ab$	$0.75\pm0.11 \ ab$
32	Isorhamnetin-C- hexoside	$0.03\pm0.01\ a$	$0.01\pm0.00\ b$	-	$0.04\pm0.01\ a$	$0.02\pm0.00\ b$	$0.01\pm0.00\ bc$
33	Naringin	$0.02\pm0.00\ \text{c}$	$\begin{array}{c} 0.03 \pm 0.01 \\ abc \end{array}$	$0.04\pm0.01\ a$	$\begin{array}{c} 0.03 \pm 0.00 \\ abc \end{array}$	$0.01\pm0.00\ bc$	$0.03\pm0.00\ ab$
34	Guaiacyl(8- <i>O</i> - 4)syrinigyl(8- 8)guaiacyl-hexoside	0.01 ± 0.00	0.03 ± 0.02 a	$0.01\pm0.00~b$	0.02 ± 0.01 ab	0.03 ± 0.00 a	$0.01\pm0.01~b$
38	Trihydroxy-methoxy- flavonol	$0.05\pm0.01\ b$	$0.06\pm0.01\ b$	$0.11\pm0.02~a$	$0.11\pm0.01~\text{a}$	$0.05\pm0.02\ b$	$0.05\pm0.01\ b$

Table 5. Concentration (mg/g dw) of (poly)phenolic compounds in fruit pulp of six cultivars

of Opuntia ficus-indica.

Id.	Compounds	FRESA	NA	NALLE	NE	NO	NT
3	Protocatechuic acid- hexoside	$0.02\pm0.00\ bc$	$0.01\pm0.00\ \text{c}$	$0.08\pm0.02\ a$	$0.02\pm0.00\ bc$	$0.02\pm0.01\ bc$	$0.03\pm0.00\ b$
6	Myricetin-hexoside	-	-	-	-	-	$0.01\pm0.00\;a$
7	Ferulic acid derivative	0.08 ± 0.02	-	-	-	-	-
9	Ferulic acid-hexoside	$0.14\pm0.03\ a$	$0.02\pm0.02\;b$	$0.06\pm0.00\ b$	$0.02\pm0.00\ b$	$0.05\pm0.01\ b$	$0.15\pm0.03\ a$
10	Guaiacyl(t8-O- 4)guaiacyl-hexoside	$0.19\pm0.02\;b$	$0.19\pm0.01\ b$	$0.10\pm0.00\;d$	$0.18\pm0.02\ bc$	$0.14\pm0.03~\text{cd}$	$0.33\pm0.02\ a$
12	Sinapic acid-hexoside	$0.10\pm0.01\ b$	$0.21\pm0.05\ b$	$1.71\pm0.36\ a$	$0.06\pm0.01\ b$	$0.06\pm0.01\ b$	$0.10\pm0.02\ b$
17	Syrinigyl(t8-O- 4)guaiacyl	$0.13\pm0.04\ ab$	$0.12\pm0.01\ b$	$0.08\pm0.01~\text{c}$	$0.07\pm0.02~\text{c}$	$0.06\pm0.01\ c$	$0.17\pm0.01\ a$
20	Isorhamnetin- rhamnose-rutinoside	$0.01\pm0.00\ a$	-	-	-	-	$0.01\pm0.00\ a$
21	Quercetin-hexoside- pentoside	$0.01\pm0.00\ a$	-	-	$0.01\pm0.00\ a$	-	-
22	Isorhamnetin derivative	-	-	$0.01\pm0.00\ a$	-	-	$0.01\pm0.00\ b$
23	Dihydrosinapic acid hexoside	-	-	$2.39\pm0.28\;a$	-	$0.12\pm0.01\ b$	-
25	Secoisolariciresinol- hexoside	-	-	0.10 ± 0.02	-	-	-
26	Isorhamnetin derivative	$0.02\pm0.00\ a$	$0.01\pm0.00\ a$	$0.02\pm0.00\ a$	0.02 ± 0.00	$0.01\pm0.00\ a$	$0.02\pm0.00\ a$
27	Quercetin-hexoside	$0.01\pm0.01\ a$	-	-	$0.01\pm0.00\ a$	-	-
29	Syringaresinol	$0.07\pm0.01\ b$	$0.02\pm0.00\ cd$	$0.13\pm0.03\ a$	$0.02\pm0.01\ d$	$0.06\pm0.01~\text{bc}$	$\begin{array}{c} 0.06 \pm 0.01 \\ bcd \end{array}$
30	Naringenin-hexoside	-	-	0.21 ± 0.04	-	-	-
31	Isorhamnetin- rutinoside	$0.02\pm0.00\ a$	$0.01\pm0.00\ a$	$0.01\pm0.01\ a$	$0.01\pm0.00\ a$	$0.01\pm0.00\ a$	$0.01\pm0.00\ a$
33	Naringin	$0.04\pm0.01\ ab$	$0.03\pm0.00\ bc$	$0.05\pm0.01\ a$	$0.01\pm0.00\ c$	$0.02\pm0.00\ bc$	$0.03\pm0.00\ ab$
34	Guaiacyl(8- <i>O</i> - 4)syrinigyl(8- 8)guaiacyl-hexoside	$0.16\pm0.03\ a$	$0.12\pm0.01\ ab$	$0.05\pm0.01\ b$	0.13 ± 0.03 ab	$0.04\pm0.01\ b$	$0.08\pm0.08\ ab$
36	Feruloyl derivative	0.96 ± 0.07 a	$0.7\pm0.14\ b$	$0.08\pm0.01~\text{c}$	$0.28\pm0.03~\text{c}$	$0.11\pm0.01~\text{c}$	1.06 ± 0.19 a
38	Trihydroxy-methoxy- flavonol	-	-	0.01 ± 0.00	-	-	-